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EDITED BY L. J. SPENCER, C.B.E., F.R.S.

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NOTE

seventh volume (1938-40) of *Mineralogical Abstracts* contains abstracts. They are roughly grouped under the following headings:

- Artificial Minerals (pp. 137, 282, 471).
- Bibliographies (p. 322).
- Chemical Crystallography (pp. 277, 480).
- Crystals (pp. 94, 420).
- Cour of Minerals (pp. 134, 525).
- Crystal-optics (pp. 247, 403).
- Economic Minerals and Ore-deposits (pp. 147, 430).
- Growth and Corrosion of Crystals (pp. 288, 531).
- History and Biography (pp. 15, 318).
- Instruments and Apparatus (pp. 253, 399).
- Mathematical Crystallography (p. 239).
- Meteorites and Tektites (pp. 61, 171, 265, 371, 534).
- Miscellaneous (pp. 52, 104, 159, 208, 255, 308, 350, 405, 459, 506, 552).
- Museums and Collections (p. 18).
- New Minerals (pp. 9, 58, 119, 168, 221, 263, 314, 369, 417, 468, 512).
- Notices of Books (pp. 1, 55, 111, 167, 219, 311, 367, 415, 467, 511).
- Physical Properties (pp. 243, 400).
- Precious Stones (pp. 124, 328, 516).
- Radomorphs (p. 500).
- Radioactivity (pp. 341, 497).
- Rock-forming Minerals and Petrology (pp. 20, 178, 292, 444).
- Topographical Mineralogy (pp. 112, 335, 502, 548).
- Ways and Crystal-structure (pp. 79, 226, 381, 487).
- Willites (p. 344).

In the preparation of the abstracts the indexing has been kept constantly in view. Many points receive a bare mention in the abstracts without their inclusion in the index. It is hoped that by this means the reader will be put in touch with the original literature. The Alphabetical Index (pp. 578-663) of AUTHORS, subjects, and Localities is preceded by a Topographical Index (pp. 559-577).

ERRATA AND ADDENDA

Mineralogical Abstracts, vol. 2.

PAGE.	LINE.	
133	3*	<i>For HO read H₂O</i>
245	13	<i>For Gazette read Gazzetta</i>
568	11*	<i>For Bohemia read Moravia</i>

Mineralogical Abstracts, vol. 3.

394	11	<i>For 2·33 read 9·33</i>
394	5*	<i>For 48·81 read 40·81</i>

Mineralogical Abstracts, vol. 6.

339	17	<i>For 10,000 read 100,000</i>
443	19	<i>For (SO₃) read (SO₄)</i>
544	7* (col. 1)	<i>For 354–6 read 254–6</i>
578	16* (col. 2)	<i>For sulparsenites read sulpharsenites</i>
585	1* (col. 1)	<i>For CuCl₂.H₂O read CuCl₂.2H₂O</i>

Mineralogical Abstracts, vol. 7.

1	6*	<i>For newberryite read newberyite</i>
10	8	<i>For научной read научной</i>
31	17*	<i>For Petrographic read Petrologic</i>
59	4	<i>For Quetana read Quetena</i>
64	13*	<i>For Davies read Davis</i>
67	15*	<i>For fuset read fused</i>
74	2*	<i>For 6–410 read 6–401</i>
75	18	<i>For Wein read Wien</i>
114	5*	<i>For gold mine read copper mine</i>
115	9	<i>For MgO 4·45 read MgO 4·54</i>
120	4	<i>For (OH)₃ read (OH)</i>
134	18	<i>For pp. 111–113 read pp. 111–114</i>
134	24	<i>For lazulite read lazurite</i>
152	12	<i>For северо-босточного read северо-восточного</i>
169	13	<i>Add β 67° to axial ratios</i>
226	2	<i>For 2V 34° 14' read 2V 13° 14'.</i>
239	7	<i>For b 10·05 read c 10·05</i>
257	5	<i>Add 1 pl.</i>
260	13	<i>For a show read show a</i>
271	16	<i>For Kazakstan read Kazakhstan</i>
282	4*	<i>For BORKY (G. B.) read BOKY (G. B.)</i>
306	6*	<i>For actinolite-schist read actinolite-schist,</i>
309	14	<i>Add pp. 67–76</i>
323	9	<i>For consists read consist</i>
361	10*	<i>For Hunyady read Hunyad</i>
394	5*	<i>For x 0·025 read x 0·25</i>
418	8*	<i>Add no. 1,</i>
446	21	<i>For CaO 21·46 read CaO 12·46</i>
451	18*	<i>For 6 pl. read 6 pls.</i>
461	2	<i>For урале read Урале</i>
490	16*	<i>For pp. 566–571 read pp. 566–576</i>

* From bottom of page.

NERALOGICAL ABSTRACTS

(Vol. 7, No. 1, March, 1938.)

Notices of Books.

ix (Alfred). *Le volcan actif de l'île de la Réunion et ses produits.* Paris (Gauthier-Villars), 1936, ix+297 pp., 1 map (1:100,000), pls.

Splendidly illustrated book gives a complete account of the active (Piton de la Fournaise) of Reunion and the lithology of its Part 1 publishes for the first time a description of the island and volcano from a manuscript of Alexis Bert dated 1791. Part 2 s observations made by the author himself in 1911. Part 3 is a punt of the rocks with very many analyses, the rocks from the volcano and those from the massif of the Piton des Neiges being separately. An appendix gives biographical notices of early rs and contemporary accounts of the activity of the volcano : 1751 and 1768. The book is full of valuable information for eanologists and petrologists. Items of mineralogical interest two new analyses of olivine; alteration of olivine by oxidation of oceanite flows producing iridescent and reddened crystals have been described in the past under the names of chusite and ; observations made in reflected light on titanomagnetite, , and haematite, with descriptions of some interesting inter- ; and pseudobrookite in the central cavities of basalt stalactites. are well crystallized in amygdules in lavas of the Piton des species listed are chabazite, analcime, faroelite, mesolite, te, apophyllite, gyrolite, and natrolite. Analyses, refractive and densities are given both for basalt-glass, of which some a form like the Pelé's hair of Hawaii, and for the residual glass of the basaltic lavas. For comparison analyses of glasses from are republished, and new analyses are given of a hair-like from Tanna in the New Hebrides and of the andesite of which glassy form. Phosphates found in a basalt cave include miner- wberryite, redondite, lecontite, brushite, and struvite. Car- in fissures are calcite, aragonite, and magnesite; an analysis of nesite (d 2.927) gave: MgO 42.46, CaO 5.28, Fe₂O₃ 0.55, S 0.83, 36, H₂O 0.67, total 100.15. Sulphur filling vesicles in a basalt ed to be of magmatic origin. Some black sands carrying mag- and ilmenite yield between 16 and 23 % TiO₂. A deposit of

silica consisting essentially of siliceous cells of grasses with some water diatom tests is named 'mascareignite'. Analysis gave: 83.07, $\text{Al}_2\text{O}_3 + \text{Fe}_2\text{O}_3$ 2.00, MgO 1.44, CaO 0.64, H_2O 13.61, total 100. The origin of the deposit is unknown.

The lavas of the active volcano belong to two main types: labradorite basalts poor in olivine, and oceanites very rich in olivine and aegirine. The latter occur as true flows. They differ from the labradorite-basalts chiefly in their higher percentage of olivine, which rises from 7 to 25 per cent. Settling of olivine during crystallization is the probable cause of this difference. Stalactites and stalagmites of basalt formed in tunnels and caves, and also the 'perimorphoses' and 'eisomorphoses' (moulds and internal casts of trees) engulfed in the lava are described. The lavas forming the cone of the active volcano are classified as aphyric basalts, porphyritic olivine-basalts, 'doleritic basalts and dolerites', ankaramites, and oceanites. In the older extinct volcano of Piton des Neiges the lavas are basaltic: in earlier flows basalts are accompanied by andesites, trachytes, and a pantellerite. The basalts belong to the same kinds as those forming the cone of the active volcano with the addition of a type carrying phenocrysts of plagioclase. Certain coarse-grained rocks occur. They are classified as periodotite, harrisite, felspar, picrite, olivine-gabbro—there is a complete passage between these. Theralites, a mixed dolerite-monzonite dike, akerite, and alkali quartz-syenite. Some of these are inclusions, others are found as boulders; some are intrusions. It appears that coarse-grained rocks can be produced in the body of a volcano at relatively shallow depths although the grain of the rocks is that generally associated with plutonic conditions. A final chapter draws comparisons between the lavas of the extinct volcano and those of the active volcano, and adds some remarks on those of Mauritius and Rodriguez. Altogether there are 73 analyses of rocks as well as a classification of the rocks and their relationships.

W. C.

NIGGLI (Paul). *Das Magma und seine Produkte unter besonderer Berücksichtigung des Einflusses der leichtflüchtigen Bestandteile. I. Physikalisch-chemische Grundlagen*. Leipzig (Akademische Verlagsgesellschaft), 1937, xi+379 pp., 276+52 figs. Price 32 RM. (about 34 RM.).

'Die leichtflüchtigen Bestandteile im Magma' was the title of a thesis by Prof. Paul Niggli published in Leipzig in 1920. This book has been frequently quoted and often discussed, but its circulation

been very limited for a work of such importance and it was not accessible in this country. The author has now embodied in a more comprehensive work his views on this important subject—role of volatile constituents in rock-magmas. The aim of this to be published in two parts, is to draw from physical chemistry, petrology, and geology all that those sciences can now offer towards the explanation of the behaviour of magmas and the formation of igneous rocks. The part now published is a thorough discussion of the contributions of physical chemistry to the problem. It is of course mainly concerned with the application of the phase rule, first to 'condensed' systems (without considering gaseous phases) and then to the more difficult problems presented by systems containing volatile constituents. Binary and ternary systems which have been studied experimentally are discussed, particular attention being paid to those involving silica. Volatile phenomena are given an important place in the treatment. Sections are devoted to the influence on the fabric and texture of aggregates, of the 'kind' of crystallization, and of the presence of volatile constituents. Homogeneous equilibria are discussed in a long chapter, which includes diagrams to show the course of crystallization in rock provinces of 'Atlantic' and 'Pacific' types. All systems discussed in the book are indexed and the data for many systems giving weight and molecular percentages are tabulated. The introduction contains a table of silicates and melting-points, and of volatile constituents of rock-magmas with melting, boiling, and ignition points and critical temperatures. In order to follow the discussions the reader needs to be thoroughly conversant with the phase rule and its applications. No more elementary treatment of such complex systems is to be expected. The main conclusions which the author endeavours to be found in short summaries. Petrologists will look forward to the second part of this book, which will bring into relation with the national sciences of geology and petrology the conclusions that have been drawn from all the available physical-chemical data. The author is one of the few living mineralogists sufficiently well equipped for this task.

W. C. S.

Лодочников (В. Н.)] Лодочников (B. N.). Серпентины и серпентиниты Ильчирские и другие и петрологические вопросы, с ними связанные.—LODOČNIKOW (Wartan N.). *Serpentine und Serpentinit der Iltschirlagerstätte und im Allgemeinen und damit verbundene petrologische Probleme*. Труд. Центр. Научно-Иссл. Геол.-Разв.

Инст. (Trans. Centr. Geol. & Prosp. Inst.), Leningrad, 1936, no.
817 pp., 50 figs., 1 map. (Russian with German summary.)

This thick volume, bewildering though it may be at the first glance, is nevertheless a remarkable achievement. The account of the serpentinite complex of Ilchir (SW. of Lake Baikal), to which some 150 pages are devoted, serves as a prelude to a long discussion on the origin of serpentinite and of ultrabasic (hyperbasic) rocks in general, with occasional digressions to other problems in petrology. Those who speak Russian will undoubtedly enjoy this rambling but wittily critical and well-documented essay (it contains no less than 1950 references!). The impression left on the reader is that all authors from Theophrastus to the present day have been summoned before the judgement seat, and that few of them have come through the ordeal unscathed.

A discussion of the nomenclature in vogue for the serpentinite group of minerals heads mainly towards criticism of the superabundance of names. The name serpentinite is suggested for a rock mainly composed of serpentinite minerals and these are catalogued as antigorite, chrysotile, bastite, and serpophite [Min. Mag. 23-637], the last being a new name suggested to cover compact resinous and microcrystalline or isotropic varieties, in place of such terms as ophite, metaxite, picrolite, &c. Other names are considered superfluous. Asbestos is reserved for the fibrous elastic variety of chrysotile, while for similar varieties of amphibole the names of amianthus or byssolite are retained. The accessory minerals of Ilchir serpentinites are magnetite, brucite, fuchsite, and carbonates. Olivine and pyroxene are found as relict minerals.

In addition to the detailed description of the serpentinite complex of Ilchir, the author gives a description of the surrounding schists, in particular the serpentinitized carbonate rocks, which in his opinion are metasomatically contact-altered limestones. Rocks similar to listwänite are common, but the term listwänite is considered ambiguous and is recommended for further use. The component minerals of these schists are quartz, serpentinite, talc, sericite, fuchsite, albite, magnetite. The carbonates are of variable composition even in a small fragment (dolomite, ankerite, breunnerite, pistomesite). A new mineral from a quartzite-schist is described under the name of eikotourmaline [Min. Mag. 23-628]. This mineral has not been found in sufficient quantity to allow a chemical analysis. It occurs in the form of needles, brownish-greenish, with a patchy distribution of colour, pleochroic, absorption $\alpha < \beta$, $\beta - \alpha = 0.030-0.032$, $1.662 > \beta > 1.658$, $\alpha \ll 1.628$, negative $2V = 37.5^\circ, 40^\circ$ (brown variety), $32^\circ, 37^\circ$ (green variety). Another mineral

the amphibole, is named antiglaucophane [Min. Mag. 23-625]. It differs from normal glaucophane in its scheme of absorption ($\alpha \geq \beta > \gamma$). The most stimulating part of the book is a discussion on the origin of peridotites and ultrabasic rocks. In a study which is both historical and critical the author boldly attacks well-established gravitational and melt-fractionation hypotheses of differentiation of the magma. He accepts the independent existence of an ultrabasic magma rich in volatiles. The concentration of volatiles during the late stages of differentiation of this magma results in the autometamorphism of the newly formed solid phase, e.g. serpentinization. This volatile enrichment during the late stages is part and parcel of the 'law of polarity'—the development of the idea of E. S. Larsen [M.A. 4-129]—the late magmatic processes are characterized by those constituents which are present in relatively small amounts in the original magma. The polar elements give rise to polar minerals which are present in the late veins in the contact-metamorphic zones. Such polar elements of ultrabasic magmas are Ca, Al, Si, Na, K, and they give rise to grossular, andradite, corundum, albite, &c. The hypothesis of desilicification as due to albite and plumbosite veins is proved to be untenable and the connection of these veins with later acid intrusions is denied. These veins are genetically connected with the ultrabasic magma itself and not with its residual products. In the same way Bowen's reaction scheme, in its narrow application to dry systems, cannot be applied to magmatic processes. The order of separation of minerals from magma is not always the same, nor can it be inferred from the degree of polymorphism of the minerals. Serpentinization is determined by various factors, such as the depth, the size and the shape of the igneous body, the velocity of intrusion, the contents of the volatiles in the magma, &c. The asbestosiform variety of serpentine, to which a separate chapter is devoted, is explained as being due to the solidification of a fibrous gel under tension and not to the infilling of open fissures. The work is well indexed according to the subject-matter, place and authors, and is provided with a good German summary (pp. 768-770).

S. I. T.

ФЕРСМАН (А. Е.) Геохимия. Том III. [Geochemistry. Vol. 3] Химтеорет [Khimteoret], Leningrad, 1937, 503 pp., 12 figs. Price 15 rub. (bound 16 rub. 50 kop.).

This third volume of this work is far more extensive than the preceding [Vol. A. 6-338] and a short abstract cannot do it justice. It is, in a

way, an amplification of the previous volumes, with the introduction of the principle of energetics, especially of the concepts evolved by author during the last two or three years [M.A. 6-469, 470, 471]. Chapter VIII (the numeration of chapters and figures is carried over from the previous volumes) the periodic law of Mendeleev in its application to geochemistry is discussed. In Chapter IX the energy of crystal lattice and the *EKs* are discussed and their application to crystal-chemistry is outlined. In Chapter X the principles of energetics are applied to the paragenesis of minerals and to various geochemical processes. A new concept—‘paragene’—is introduced. Paragene function depending on crystal-chemical constants and energy index. It is suggested that, when accurately determined, it will serve to fix the relative position of compounds (minerals) in the paragenetic successions. In Chapter XI the principles of energetics are applied to the classification [M.A. 6-286] of geospheres and cosmo-spheres. The clarkes of various systems are shown to represent energetic stability levels. The main object of the present volume is to give an outline of geochemistry based on a small number of independent variables and governed by laws of energetics. The book is provided with separate appendixes to each chapter and three appendixes giving the list of terms and symbols used, a list of chemical elements with their constants, and a bibliography.

S. I.

[FERSMAN (A. E.) & SHUBNIKOVA (O. M.)] Ферсман (А. Е.) и Шубникова (О. М.). Спутник геохимика и минералога. [*Geochemist's mineralogist's companion.*] Moscow & Leningrad (Lomonosov Inst., Acad. Sci.), 1937, 415 pp., 1 pl., 48 figs. Price 17 rub. 50.

This is a useful book of reference, or vade-mecum, giving, mostly in tabular form, a vast collection of facts and data relating to minerals. The material has been compiled by several contributors under the guidance of A. E. Fersman with Mrs. O. M. Shubnikov as responsible editor. Different sections are devoted to physics and chemistry, crystallography, crystal-chemistry and isomorphism, geochemical processes, mineralogy, petrography, geology, literature, and weights and measures. L. J.

[LEVINSON-LESSING (F. Y.) & STRUVE (E. A.)] Левинсон-Лесинг (Ф. Ю.) и Струве (Э. А.). Петрографический Словарь. [*Petrographical lexicon.*] 2nd edit., Leningrad-Moscow, 1937, 416 pp.

The original ‘Petrographisches Lexikon’ of Loewinson-Lessing was written in German and published in 1893-94 contained 3329 items. A Fr.

was issued by the International Geological Congress in 1901, and a Russian edition appeared in 1933. The present second Russian edition contains 5858 items, including the names of rocks, the names of their species, and textural, structural, and other terms in use or obsolete. Every term has a reference to the literature. The second edition's work includes many modal analyses from Tröger's 'Kompendium' [M.A. 6-97].

S. I. T.

G (W. L.). *Atomic structure of minerals*. London (Oxford University Press), 1937, xiii+292 pp., 144 figs (some on 9 pls.). Price 18s. and Cornell Univ. Press, Ithaca, N.Y., 1937, Price \$3.75.]

This fifteenth volume of the series 'The George Fisher Baker non-resident lectureship in chemistry at Cornell University' [M.A. 6-1] does not contain the substance of the lectures given there by the author in

That material had been already published in his book 'The crystalline state' [M.A. 5-385]. Instead, the opportunity has been taken of presenting to mineralogists a useful summary of the crystal-structures of all minerals (arranged under species in the order: elements, oxides . . . silicates) that had been worked out up to the end of 1936. Readable descriptions of the structures are accompanied by references to the literature and many new drawings, mostly two-dimensional, showing clearly the pattern of the structure. Part I (pp. 3-40) gives general accounts of the geometry of crystal patterns, methods of X-ray analysis, and the general principles of mineral structures. This is the stereochemistry of minerals in which there are no molecules, but repetition of a pattern throughout the crystal. The isomorphous nature of end-members with a definite chemical formula 'is based on a principle which is fundamentally wrong' (p. 37). L. J. S.

ELLER (W. R.). *The analytical chemistry of tantalum and niobium: the analysis of their minerals and the application of tannin in gravimetric analysis*. With a foreword by G. Roche LYNCH. London (Chapman & Hall), 1937, xvi+198 pp. Price 21s.

This valuable treatise on a field of chemical analysis in which reliable and accurate methods were for long quite lacking is for the most part a condensed and collated summary of the results published by the author and several collaborators in a series of 33 papers, mostly in 'The Analyst'. It will be indispensable to any one attempting the analysis of minerals in which tantalum or niobium is present, even in small amount, while many of the new methods described will probably inspire

advance in other fields of analytical chemistry. Both qualitative and quantitative methods are described for the detection and estimation of niobium and tantalum in admixture with large or small amounts of most of the other elements. The new methods are largely based on the use of tannin, which serves both as a specific and as a group reagent under proper conditions. Details are given for the preparation of pure niobic and tantalic oxides.

M. H. H.

DAVISON (E. H.). *Field tests for minerals*. London (Chapman & Hall), 1937, viii+60 pp., 12 pls., 3 text-figs. Price 7s. 6d.

Clear and concise details for blowpipe tests, spot tests with organic reagents, and a few micro-chemical tests are condensed into 17 pages. Determinative tables based on lustre, hardness, and streak occupy 26 pages. The common Cornish mineral mispickel has been overlooked, and there are numerous misprints. The plates give excellent photographic reproductions of mineral specimens illustrating different types of structure, but they have little bearing on the text. The whole amounts to a single chapter in a text-book of mineralogy, and the price is much too high. L. J. S.

HOLMES (Arthur). *The age of the earth*. London (Thomas Nelson & Sons), 1937, x+263 pp., 4 figs. Price 1s. 6d.

The first edition of 1927 [M.A. 4 113] was several times reprinted, and this second edition has been entirely rewritten and much enlarged. It gives an excellent and readable account of the various methods of estimating geological time. The chapters dealing with the methods based on radioactivity give full technical details, including age formulae and tables of all the recent analyses of radioactive minerals. The author then wanders off with meteorites into space beyond the Milky Way, and concludes that all the trouble started about 2,000,000,000 B.C. A bibliography of recent scientific papers is added. The attractive little volume is well printed in clear type on good paper and is nicely bound. The author and publishers are to be congratulated on producing a speculative scientific treatise at the very low price of eighteen pence. L. J. S.

BERINGER (Carl Chr.). *Geologisches Wörterbuch. Erklärung der geologischen Fachausdrücke für Geologen, Paläontologen, Mineralogen, Ingenieure, Geographen, Bodenkundler, Studierende und alle Freunde der Geologie*. Stuttgart (Ferdinand Enke), 1937, vii+126 pp., 51 ff. Price RM. 5.60 (bound 6.90).

This is a useful little dictionary giving the meanings and derivations with concise explanations of a large number of geological terms. The

mostly to physical and structural geology. There are a few
al petrological terms, but none for mineralogy. The author of a
is in many cases stated, but there are no references to the litera-

Names of geological formations are collected in a 4-page table
alphabetical index at the end of the volume. L. J. S.

New Minerals.

ЗНИКОВА (О. М.)] Шубникова (О. М.). Новые минералы, откры-
тие в 1933–1934 гг. [New minerals discovered in the years 1933–34.]
Труды Ломоносов. Инст. Акад. Наук СССР (Trans. Lomonosov
Inst. Acad. Sci. U.S.S.R.), 1936, no. 7, pp. 307–339.

This is supplementary to the report on new minerals for the years
1932 [M.A. 6–51]. It includes also many old minerals for which new
appeared in the literature of 1933–34. L. J. S.

ЕФРЕМОВ (Н. Е.)] Ефремов (Н. Е.). Каракашит—новый минерал—
EFREMOV (N. E.). Karachaite—a new mineral. Изв. Акад. Наук
СССР, Отд. Мат. Ест. Наук. Сер. Геол. (Bull. Acad. Sci. U.R.S.S.,
Cl. Sci. Math. Nat., Sér. Géol.), 1936, pp. 921–928. (Russian with
English summary.)

Karachaite, a new mineral from the asbestos deposit of Shaman-
gen, Karachai, NW. Caucasus, is described. It is an asbestosiform
variety of chrysotile serpentine, sp. gr. about 2·20, yellowish, positive
refraction, $\alpha' 1\cdot542$, $\gamma' 1\cdot546$. Analysis by V. Krylova gave: SiO_2 47·12,
 Al_2O_3 2·82, Fe_2O_3 2·58, MgO 30·90, CaO 3·20, H_2O —13·48. H_2O =
100·86, corresponding to $\text{MgO} \cdot \text{SiO}_2 \cdot \text{H}_2\text{O}$. The relations of this
mineral to the other members of the same group is discussed.

S. I. T.

СЕРДЮЧЕНКО (Д. Р.)] Сердюченко (Д. Р.). Пикроамозит—новый
минерал.—SERDUČENKO (D. R.) [sic]. Picroamosite—a new mineral.
Изв. Акад. Наук СССР, Отд. Мат. Ест. Наук. Сер. Геол. (Bull.
Acad. Sci. U.R.S.S., Cl. Sci. Math. Nat., Sér. Géol.), 1936, pp. 689–
96. (Russian with English summary.)

Picroamosite, a new variety of orthorhombic amphibole, was found
in the region of the river Malaya Laba, northern Caucasus. It is fibrous,
greenish-grey, almost colourless in section, slightly pleochroic,
refraction straight, $\alpha 1\cdot626$, $\beta 1\cdot638$, $\gamma 1\cdot651$, $2V 87^\circ$, positive, dispersion
 $v > r < v$. Analysis gave: SiO_2 55·90, TiO_2 nil, Al_2O_3 1·13, Fe_2O_3
 FeO 0·68, MnO 0·49, MgO 29·26, CaO 0·94, alkalis nil, H_2O —3·20 =

99·88. Two isomorphous series are postulated: ferroanthophyllite-amosite-gedrite and magnesioanthophyllite-picroamosite-ferrigedrite. The genesis of picroamosite is related to the action of granitic magmas on the serpentine-rocks of the region.

S. I. T.

[ЧИРВИНСКИЙ (П. Н.)] Чирвинский (П. Н.). Фошалласит из Хибинской тундры.—TSCHIRWINSKY (Peter). *Foschallasit aus der Chibinischen tundra*. Академику В. И. Вернадскому к пятидесятилетию научной и педагогической деятельности. Акад. Наук СССР [Vernadsky jubilee volume, Acad. Sci. U.S.S.R.], 1936, vol. 1, pp. 757–763, 3 figs. (Russian with German summary.)

This mineral occurs as snow-white veinlets associated with calcite and mesolite in the lovchorrite mine on Yukspor Mt. It is scaly, sometimes as spheroidal aggregates; sp. gr. 2·5, H. 2½–3. The minute scales have a perfect tabular cleavage with pearly lustre; $Bx_a \perp$ cleavage, $2V\ 12-15^\circ$, negative, $\alpha\ 1\cdot535$, $\beta\ 1\cdot542$, $\gamma\ 1\cdot549$, $\gamma - \alpha\ 0\cdot018$, $\gamma - \beta\ 0\cdot001$. Analysis: I. D. Borneman-Starynkevich gave $SiO_2\ 32\cdot65$, $R_2O_3\ 1\cdot89$, $CaO\ 45\cdot00$, $Na_2O\ 0\cdot40$, $H_2O\ 0\cdot16$, ign. 16·66 = 97·21; formula $3CaO \cdot 2SiO_2 \cdot 3H_2O$. The mineral is related to foshagite [M.A. 2–520] and centrallassite, from which it is named *foshallassite*.

L. J. S.

QUENSEL (Percy). *Minerals of the Varuträsk pegmatite. V. Manganapatite and manganvoelckerite*. Geol. För. Förh. Stockholm, 1936, vol. 59, pp. 257–261. [Cf. M.A. 6–485–7.]

Bluish-green prisms of manganapatite, associated with albite, have sp. gr. 3·22, $\omega\ 1\cdot6459$, $\epsilon\ 1\cdot6411$, and gave anal. I. A paler blue apatite found as remnants at the bottom of pits on the weathered surface of the rock, has $\omega\ 1\cdot6402$, $\epsilon\ 1\cdot6365$ and gave II, in which CaF_2 is largely replaced by CaO . This is called manganvoelckerite. It sometimes forms a zone around manganapatite, and the suggestion is that the supply of fluorine was cut off during the later stages of growth.

P_2O_5	Fe_2O_3	FeO	MnO	MgO	CaO	$H_2O +$	$H_2O -$	F.	Insol.	To
I. 41·50	—	0·26	5·32	0·04	50·31	0·25	0·03	3·41	0·35	99
II. 41·84	0·35	nil	4·31	0·10	52·00	0·17	0·04	1·43	0·18	100
II, also	Al_2O_3	0·15	Na_2O	0·03	K_2O	0·07	Li_2O	0·01	Cl	0·02

L. J. S.

DE LEENHEER (L.). *Sur l'hydroténorite, minéral nouveau; sur la ténorite et sur l'identité de ce dernier minéral avec la mélanochalcite*. Bull. Soc. Belge Géol., 1937, vol. 47, 3 pls., 5 text-figs.

Hydroténorite is an amorphous black mineral banded with chrysocolla from Étoile du Congo mine, Elisabethville, Katanga. Polished section

-ray photographs show indications of crystalline structure; sp. gr. H. $3\frac{1}{2}$. Analysis gave CuO 80.87, Co_2O_3 0.90, SiO_2 8.04, H_2O 99.58; deducting chrysocolla, the formula is deduced as H_2O . At 43°C . there is a loss of 4.26 % H_2O , at 120° of 5.76 %, 90° 9.75 %, and at 1000° oxygen is lost. Tenorite, occurring as an intermediate product of cuprite at Likasi, Katanga, has much the same appearance, but gave more definite indications of crystalline structure; sp. gr. 4.984, H. $3\frac{1}{2}$. Analysis, CuO 87.13, SiO_2 5.17, CO_2 1.02, H_2O 99.30, is interpreted as a mixture of CuO, chrysocolla, and tenorite, as with melanochalcite and pitchy copper ore [M.A. 1-263, 1937].

L. J. S.

(R.). *The minerals formed on burning heaps in the coal basin of Kladno*. Bull. Internat. Acad. Sci. Bohême, 1937, preprint 7 pp., 1 pl., 9 text-figs.

Minerals described from the burning waste heaps [M.A. 6-357] are β -sulphur, γ - and red selenium, tschermigite (čermíkite), epsomite, hydrite, pickeringite, halotrichite, alunogen, mascagnine, sal-ammoniac. Lapparentite [M.A. 5-390, 6-149] as chalk-white masses gave α 1.470, β 1.484, sp. gr. 1.892; SO_3 35.51, Al_2O_3 22.07, MgO 0.52, Fe_2O_3 1.09, H_2O 40.69 = 99.88; 1 mol. H_2O is lost at 47°C ., and seven percent at 120° ; formula $\text{Al}_2\text{O}_3 \cdot 2\text{SO}_3 \cdot 10\text{H}_2\text{O} = \text{Al}_2(\text{OH})_2(\text{SO}_4)_2 \cdot 9\text{H}_2\text{O}$. Letovicite has α 1.511, β 1.543, γ 1.583; SO_3 40.09, Al_2O_3 2.94, Fe_2O_3 1.54, FeO 0.54, MnO 0.10, MgO 3.04, CaO 0.67, Na_2O 0.53, K_2O 0.27, H_2O 33.42 = 100.48. Letovicite [M.A. 5-145] gave α 1.501, β 1.516, γ 1.525, sp. gr. 1.804; SO_3 63.92, NH_3 20.19, agreeing with $(\text{NH}_4)_3\text{SO}_4 \cdot 2\text{H}_2\text{O}$. Kratochvilite as pale bluish-violet or greenish pearly scales has α 1.557, β 1.725, sp. gr. 1.19, m.p. 113°C .; analysis, C 92.44, H 7.28, H_2O approximately with $\text{C}_{13}\text{H}_{10}$ (= fluorene, M.A. 6-357). Another organic compound is probably an anthracene derivative. L. J. S.

(B. Rama) & RAO (L. Rama). *On "bidalotite" a new orthorhombic pyroxene derived from cordierite*. Proc. Indian Acad. Sci., Sect. B, 1937, vol. 5, pp. 290-296, 3 pls.

Biotite-cordierite-hypersthene-granulite near Bidaloti in Mysore is a metamorphosed sediment intruded by granite. It contains a lilac-red orthorhombic pyroxene with striking pleochroism, α pale yellow to colourless, β and γ lilac or purple; α 1.656, β 1.667, γ 1.672, $2V$ 57°, negative. Two of the four analyses are SiO_2 53.16, Al_2O_3 1.35 (1.73), TiO_2 1.35 (1.73), Al_2O_3 10.55 (4.90), Fe_2O_3 4.30 (3.43), FeO 17.10

(15.43), MgO 11.95 (18.19), Na₂O trace (1.38), K₂O nil (trace), H₂O + 2 (1.80), total 100.41 (99.22), sp. gr. 3.20 (3.22). As small grains and plates, this mineral, named bidalotite, is an alteration product of cordierite. The pleochroic haloes in the cordierite persist in the new mineral. The associated hypersthene has the usual pleochroism, α bright pinkish-red, γ bluish-green, n 1.72; and the new mineral also differs from this in containing alumina.

L. J. S.

MEIXNER (Heinz) & PILLEWIZER (Wolf). *Über Minerale, die teils im Schrifttum, teils in Sammlungen als "Keramohalit" bezeichnet werden (Bosjemanit von Terlan in Südtirol, Eisenpickingerit [sic] von Dienten, Pickingerit [sic] von Mitterberg in Salzburg und ein Halotrichitvorkommen).* Zentr. Min., Abt. A, 1937, pp. 263-270.

The scaly habit and higher melting-point (113-114° C.) of keramohalite (= alunogen) distinguish it from the 'hair-salts' of the halotrichite group, R"Al₂(SO₄)₄.23H₂O. Analyses of 'hair-salts' from Terlan in Trentino and from Dienten and Mitterberg are interpreted as mixtures of minerals of the halotrichite group sometimes with epsomite, and the names magnesium-apjohnite [= bushmanite = bosjemanite], 'Manganickingerit', and 'Eisenpickingerit' [i.e. pickeringite!] are suggested.

L. J. S.

DITTLER (E.) & KIRNBAUER (F.). *Die Kaolinlagerstätte von Tornielli (Mittelitalien).* Zeits. Prakt. Geol., 1937, vol. 45, pp. 117-124
4 figs.

A 20 m. vein of kaolin in quartz-trachyte is worked at Tornielli 56 km. S. of Siena. A network of veinlets in the kaolin is filled with yellowish or grey clay mineral which is completely amorphous, showing no X-ray pattern; sp. gr. 2.432, H. 2, n 1.535-1.536. Analysis of material air dried at 18° gave SiO₂ 33.45, TiO₂ trace, Al₂O₃ 30.46, Fe₂O₃ 0.2, MgO 0.02, CaO 0.01, Na₂O 0.03, K₂O 0.04, P₂O₅ 0.48, H₂O + 15.6 H₂O (110°) 1.37, H₂O (over H₂SO₄) 18.43 - 100.24; ratios Al₂O₃:SiO₂:H₂O = 1:1.88:6.58; formula Al₂Si₂O₅(OH)₄.H₂O. Dehydration curves are given. The mineral is an amorphous form of halloysite and is named *torniellite*.

L. J. S.

GRIM (R. E.), BRAY (R. H.), & BRADLEY (W. F.). *The mica in argillaceous sediments.* Amer. Min., 1937, vol. 22, pp. 813-829, 2 figs.

A micaceous constituent, in particles usually less than 1 μ , separated by aqueous suspension from clays and shales from Illinois gave for the

samples of the fine colloidal fractions: I, Ordovician shale near Calhoun Co.; II, Pennsylvanian underclay near Fithian, Ver Co. The approximate formula $2\text{K}_2\text{O} \cdot 3\text{MO} \cdot 8\text{R}_2\text{O}_3 \cdot 24\text{SiO}_2 \cdot 12\text{H}_2\text{O}$ less K_2O and more H_2O than in muscovite. Optical and X-ray and dehydration curves also show certain differences from muscovite and sericite. The name *illite* is suggested as a general term for constituent of argillaceous sediments. [Cf. pholidoide and phyllite, *Min. Mag.* **24**-620-1.]

	TiO_2 .	Al_2O_3 .	Fe_2O_3 .	FeO .	MgO .	CaO .	Na_2O .	K_2O .	Ign. (+110°).
10	0.50	25.12	5.12	1.52	3.93	0.35	0.05	6.93	6.82
22	0.53	25.19	4.59	1.70	2.84	0.16	0.17	6.09	7.49
al.	$\text{H}_2\text{O}+$.	$\text{H}_2\text{O}-$.	$\text{SiO}_2 : \text{R}_2\text{O}_3$.		$\text{SiO}_2 : \text{Al}_2\text{O}_3$.	γ .	$\gamma - \alpha$.	2V.	
44	7.18	1.90		3.00		3.39	1.598	0.033	5°
70	7.14	1.45		3.02		3.36	1.588	0.033	5

L. J. S.

v (Mark C.). *Lopezite, a new mineral*. Amer. Min., 1937, vol. 22, p. 929-930.

is potassium dichromate ($\text{K}_2\text{Cr}_2\text{O}_7$) and occurs as minute orange-yellow grains on tarapacaite (K_2CrO_4) and dietzeite in vugs in the soda-silicate deposits of Tocopilla and Iquique in Chile. Implanted on it are small crystals of ulexite. It gave $\alpha 1.714$, $\beta 1.732$, $\gamma 1.805$, and microchemical tests in agreement with the artificial salt. L. J. S.

ON (Dwight M.). *Woodhouseite, a new mineral of the beudantite group*. Amer. Min., 1937, vol. 22, pp. 939-948. 4 figs.

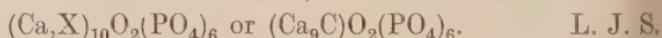
ell, colourless to flesh-coloured crystals occur in vugs in quartz in the andalusite deposits [M.A. 5-288], together with augelite [6-288], lazulite, &c., at White Mountain, Mono Co., California. They are rhombohedral, $a:c = 1:1.170$ ($\alpha 91^\circ 42'$), with perfect basal cleavage and cube-like habit or platy on (111). Sp. gr. 3.012, H. $4\frac{1}{2}$; optically uniaxial, $\omega 1.636$, $\epsilon 1.647$, but larger crystals show biaxial effects with 2V up to 20°. Analysis by A. Rautenberg, P_2O_5 18.13, 7.59, SiO_2 0.30, Al_2O_3 36.63, MgO 0.11, CaO 12.31, SrO 0.25, 0.00, Na_2O 0.08, K_2O 0.02, $\text{H}_2\text{O}+$ 13.25, $\text{H}_2\text{O}-$ 0.20 = 99.87, $2\text{CaO} \cdot 3\text{Al}_2\text{O}_3 \cdot \text{P}_2\text{O}_5 \cdot 2\text{SO}_3 \cdot 6\text{H}_2\text{O}$. This is the composition originally assigned to svanbergite, which was afterwards found to contain SrO instead of CaO [Min. Mag. 22-252]. The relations of the alunite-beudantite group are discussed. L. J. S.

McCONNELL (Duncan). *The substitution of SiO₄- and SO₄-groups for PO₄-groups in the apatite structure; ellestadite, the end-member.* Amer. Min., 1937, vol. 22, pp. 977-986, 2 figs.

A mineral resembling wilkeite, occurring as pale rose stringers with wollastonite, idocrase, and diopside at Crestmore, California, approaches the end-member of the apatite group with P₂O₅ almost wholly replaced by SO₃ and SiO₂. Analysis by R. B. Ellestad gave P₂O₅ 3·06, SO₃ 20·6, SiO₂ 17·31, CO₂ 0·61, Al₂O₃ 0·13, Fe₂O₃ 0·22, MnO 0·01, MgO 0·4, CaO 55·18, Cl 1·64, F 0·57, H₂O+ 0·53, H₂O- 0·10 = 100·52. The following data for this mineral are compared with those for two specimens of wilkeite from Crestmore and fluorapatite from Farada, Ontario [M.A. 5-480].

	<i>a.</i>	<i>c.</i>	<i>c/a.</i>	Sp. gr.	<i>ω.</i>	<i>ε.</i>	P ₂ O ₅		
Ellestadite	9·53 Å.	6·91	0·7251	3·068	1·655	1·650	3·06
Wilkeite	9·48	6·91	0·7289	3·120	1·650	1·646	14·4
"	9·40	6·89	0·7330	3·157	1·640	1·636	32·2
Fluor-apatite	9·36	6·88	0·7350	3·18	1·634	1·630	41·30

Si, S, and C replace P ions in the structure. A deficiency in F requires the presence of the voelckerite molecule, which is written as



HURLBUT (Cornelius S., Jr.) & TAYLOR (Ralph E.). *Hilgardite, a new mineral species, from Choctaw salt dome, Louisiana.* Amer. Min., 1937, vol. 22, pp. 1052-1057, 4 figs.

The sandy insoluble residue accumulating at the bottom of a brine well in rock-salt consists mainly of anhydrite together with dolomite, magnesite, hilgardite, boracite, danburite, &c. The colourless glassy crystals, $\frac{1}{2}$ - $1\frac{1}{2}$ cm. across, of hilgardite have the form of triangular plates on (010), being monoclinic-domatic with the dominant forms *b* (010), *p* (111), *M* ($\bar{1}10$), *E* (0 $\bar{1}\bar{1}$), and other small faces; *a:b:c* = 1·0147:1·05585, β = 90° 0'. Cleavage (010) perfect, (100) good. Sp. 2·71, H. 5, α 1·630, β 1·636, γ 1·664, 2V 35°, positive, β = *b*, γ :*c* = 1·0147. X-ray measurements gave *a* 11·35, *b* 11·12, *c* 6·20 Å., space-group $C_{\bar{1}}^1$ or $C_{\bar{2}}^2$. Analysis by F. A. Gonyer, B₂O₃ 49·18, Cl 10·37, CaO 34·1, H₂O 6·31, insol. 1·89, total (less O for Cl) 99·82, gives Ca₈B₁₈O₃₃(OH)₄H₂O, which occupies the unit cell. L. J. S.

PALACHE (Charles), BAUER (L. H.), and BERMAN (Harry). *Yeatmanite, a new mineral, and sarkinite from Franklin, New Jersey.* Amer. Min., 1937, vol. 22, no. 12, pt. 2, p. [11]. (Abstract.)

Clove-brown plates of yeatmanite embedded in willemite are triclinic (pseudo-orthorhombic) with perfect cleavage (100), twinned on (021)

multiple twinning on (010). X-ray measurements give $a:b:c = 1:0.4775$, $\alpha 103^\circ 49'$, $\beta 101^\circ 45'$, $\gamma 87^\circ 12'$. Sp. gr. 4.80, H. 4; $\beta 1.905$, $\gamma 1.910$, $2V 49^\circ$, negative. Composition $(\text{Mn}, \text{Zn})_{16}\text{O}_{29}$.

L. J. S.

LOCK (M. A.). *Goldschmidtine, a new antimonide of silver*. Amer. Min., 1937, vol. 22, no. 12, pt. 2, pp. [11]–[12]. (Abstract.)

This mineral, distinct from dyscrasite (Ag_3Sb), was found with native silver, ruby-silver, and galena on a specimen from Andreasberg, Harz. It is white to lead-grey, orthorhombic crystals are pseudo-hexagonal and twinned on (110), with $a:b:c = 0.6312:1:0.6860$. Sp. gr. 6.83, b_{10} . The base-centred unit cell, $a 7.75$, $b 12.32$, $c 8.42$ Å., contains $\text{Ag}_{10}\text{Sb}_{10}$. Analysis by F. A. Gonyer gave Ag 64.78, Sb 35.01, S 0.06 =

L. J. S.

History and Biography.

WALKS (Mary Elvira). *The discovery of the elements*. 3rd edition, Easton, Pa. (Mack Printing Co.), 1935, 371 pp. Price \$3.00.

The title of this book, which is a collective reprint of a series of articles in the 'Journal of Chemical Education' [M.A. 6–249], is rather misleading; it is not so much an account of the discovery of the elements as a collection of biographical sketches of the discoverers, from Brandt and Kunckel to Coster and von Hevesy. Embellished with 195 portraits (it does not mean that the 92 elements had 195 discoverers; there are less than seven portraits of Berzelius) and 95 other illustrations, including several facsimiles of letters, this book may well be found of interest for a leisure hour. Too literal translation from French or German occasionally has amusing effects.

M. H. H.

WALKS (Mary Elvira). *The discovery of tellurium*. Journ. Chem. Education (Amer. Chem. Soc.), 1935, vol. 12, pp. 403–409, 6 figs. A detailed account of the discovery of tellurium in Nagyág gold ores [tellurite and nagyagite], with translations of correspondence between Roth and Kitaibel, and portraits of Ignaz von Born (1742–1791) and Paul Kitaibel (1757–1817).

M. H. H.

WEINER (Peter). *Historische Notiz zur stereographischen Projektion*. Sitzschr. Akad. Wiss. Berlin, Abt. Min., 1936, pp. 251–252.

The principle of the stereographic projection is mentioned in the astronomical works of Ptolemy (about A.D. 150).

L. J. S.

FLETT (Sir John Smith). *The first hundred years of the Geological Survey of Great Britain.* London (His Majesty's Stationery Office), 1933. 280 pp., 14 pls. Price 7s. 6d.

A chapter on the antecedents that led to the founding of the Survey is followed by separate chapters dealing with the progress under each of the successive directors: H. T. De la Beche (1835–55), R. I. Murchison (1855–71), A. C. Ramsay (1871–81), A. Geikie (1882–1901), J. J. D. Teall (1901–14), A. Strahan (1914–20), and J. S. Flett (1920–35). An account is given of the opening ceremony of the new museum and offices in 1935; and an appendix gives a list, with biographical details, of 291 geologists who have been on the staff. L. J. S.

LAITAKARI (Aarne). *Suomen Geologisen Seura historiikki 1886–1936.*

With an English summary: *The history of the Geological Society of Finland 1886–1936.* Bull. Comm. Géol. Finlande, 1936, no. 11 (= Compt. Rend. Soc. Géol. Finlande, no. 9), pp. 5–64, 5 p. 2 text-figs.

With this history of the society, founded in 1886, are several appendices giving lists of members, lectures, papers published, &c. Portraits are given of F. J. Wiik (1839–1909) the first president of the society, Nils Nordenskiöld (1792–1866), K. A. Moberg (1840–1901), J. Sederholm (1863–1934), W. Ramsay (1865–1928), K. H. N. Gyllin (1858–1886), and others. Together with 27 papers by various authors this forms the jubilee volume of the society. L. J. S.

BAYLEY (W. S.). *Mineralogy's contribution to the other sciences and industry.* Amer. Min., 1937, vol. 22, pp. 147–168.

Presidential address giving an historical review. L. J. S.

SPENCER (L. J.). *Some mineral names.* Amer. Min., 1937, vol. 22, pp. 682–685.

Examples are given of the changes in meaning which the names of minerals may undergo in the course of time (e.g. alcohol, sapphirine, siderite, &c.), and of the confusion arising when the same name is applied to different minerals. L. J. S.

HILLER (Joh. Erich). *Boëce de Boodt, précurseur de la minéralogie moderne. Boetius de Boodt. A turning-point in the history of mineralogy.* Ann. Guébhard-Séverine, Neuchâtel, 1936, vol. 1, pp. 74–80 (French), pp. 80–81 (outline in basic English).

Anselm Boethius de Boodt (1550–1632) was born at Bruges and died there on June 21, 1632. He was physician to Rudolph II at Prague.

he wrote his 'Historia gemmarum et lapidum', published at Ver in 1609 (later Latin editions in 1636 and 1647, and French editions in 1644 and 1649). This dealt not only with gemstones and supposed medicinal virtues and counterfeits, but described 647 different kinds of gems, minerals, fossils, and stones, and it is really an treatise on mineralogy.

L. J. S.

(P.). *Domenico Guglielmini e la cristallografia*. Periodico Min. Roma, 1937, vol. 8, pp. 163-175.

account is given of the crystallographic work of D. Guglielmini (1661-1710), who wrote 'Riflessioni filosofiche dedotte dalle figure de' (Bologna, 1688; reprinted Padova, 1706) and 'De salibus dissertatione polaris physico-medico-mechanica' (Venetia, 1705). He observed each salt had a form of its own with constant angles, and he elaborates views on the structure of crystals.

L. J. S.

(Walther). *Abraham Gottlob Werner (1749-1817) in Schilddungen bedeutender Zeitgenossen*. Mitteilungen des Roland. Verein für Sippenforschung und Wappenkunde, Dresden, 1936, vol. 21, pp. 54-60.

Abraham Gottlob Werner in Darstellungen der bildenden Künste. Ibid., pp. 61-67, 5 pls.

(LZ (-) & WENSCH (Kurt). *Die Ahnen Abraham Gottlob Werners*. Ibid., pp. 67-68.

preciations by contemporary authors, including Goethe. Forty reproductions are given of portraits, busts, and medals, several of which are copies varying somewhat from the originals. His only known ancestors are father, mother, and maternal grandfather.

L. J. S.

(A. W.). *The pioneers in South African geology and their work*. Trans. Geol. Soc. South Africa, 1937, annexure to vol. 39, iii + 139 pp, 7 pls.

Biographical notes and a review of their work are given of many geologists from the eighteenth century up to the time of the foundation of the Geological Society of South Africa in 1895. Portraits are given of G. Atherstone (1814-98), who determined the first diamond found in South Africa, E. J. Dunn (1844-1937), and G. A. F. Molengraaff.

L. J. S.

ERDMANNSDÖRFFER (O. H.). *Carl Heinrich Ferdinand Rosenbusch. Zu seinem 100. Geburtstage 24. Juni 1936.* Sitz.-ber. Heidelberg Akad. Wiss., Math-nat. Kl., 1936, Abh. 3, 12 pp.
An appreciation of the influence of H. Rosenbusch [1836–1914] on the development of petrography. L. J. S.

Museums and Collections.

The student's index to the collection of minerals. British Museum: Natural History. 27th edit., London, 1936, 39 pp., 3 plans on wrapped paper. Price 1s.

This shows an increase of 6 pages on the 26th edition of 1922 [M.A. 2–6]. It lists 3864 names of species, varieties, and synonyms of minerals represented in the collection, with references to the cases where they are kept. Chemical names are included, but these are not clearly distinguished from compound mineral names. Iron sulphide is sulphide of iron, but iron pyrites (better iron-pyrites) is not pyrites of iron.

L. J. S.

A guide to the Mineral Gallery. British Museum: Natural History. 14th edit., London, 1937, viii+60 pp., 3 pls., 3 plans on wrapped paper. Price 6d.

This is an extension of the penny guide [M.A. 1–145, 2–6] which first appeared in 1889, rewritten and with the addition of three plates. It is popular in character, being intended for the ordinary visitor. There are several minor errors and inconsistencies.

L. J. S.

WALKER (T. L.) & PARSONS (A. L.). *The Royal Ontario Museum Mineralogy.* Museums Journ., London, 1936, vol. 35, pp. 412–413, 1 pl., 1 text-fig.

Repetition of a previous paper [M.A. 5–395]. The collection has been largely added to by collecting expeditions by members of the staff to Canadian mineral localities, and by using this material, of which a large stock is available, for exchanges with other museums.

L. J. S.

Catalogue of the crystal models and mineral specimens in the students' five year series. Department of Mineralogy and Petrology, Cambridge, 1934, 33 pp. (interleaved). Price 1s.

Brief descriptions of 55 crystal models and 237 mineral specimens.

L. J. S.

Notes on the museums of Ceylon, British Malaya, the West Indies, etc. London (Museums Assoc.), 1933, 58 pp.

History of museums in Ceylon, British Malaya, Hong Kong, Sarawak, British North Borneo, Fiji, the West Indies, British Guiana. London (Museums Assoc.), 1934, 67 pp. Price 5s. with Report.

THAM (S. F.). The museums of India. Part I, *A general survey.* Part II, *Directory of museums and art galleries.* London (Museums Assoc.), 1936, 229 pp., 25 pls. Price 5s.

These vols. 5 and 6 complete the series on the museums of the British Empire [M.A. 6-394].

L. J. S.

HINLEY (W. C.). Directory of American mineral and gem collectors and their collections. 1st edition, 1936, 22 pp. Price 10c.

Under towns, arranged alphabetically, in each of the States of U.S.A., names and addresses are given of some 2,000 private collectors, amateur dealers, and public museums. The pamphlet is published by the author but without any address or place of publication. The author's address, however, appears in the directory under Peoria, Illinois.

L. J. S.

ENBURG (Edmund). Internationaler Geologen- und Mineralogen-Kalender 1937. Herausgegeben von der Deutschen Geologischen Gesellschaft. Stuttgart (F. Enke), 1937, xvi+558 pp. Price RM. 10.

This edition shows an increase of 192 pages on the last issue for 1933-[M.A. 5-198]. The address-list (376 pp.) gives over 10,000 names of geologists, mineralogists, &c. Surveys, universities and colleges with more important museums, and societies (sections II-IV in previous issues) are now collected together under countries and towns.

L. J. S.

MUNISTER (F. A.). The preservation of minerals and meteorites. Museums Journ., London, 1937, vol. 36, pp. 465-476.

A long list is given of the localities of those specimens of pyrite and malachite in the British Museum collection of minerals that show signs of decomposition. This can be arrested by the treatment previously described [M.A. 5-396]. Lists are given of deliquescent and efflorescent minerals, and those altered by exposure to light or changes in temperature.

Colour changes in certain minerals when exposed to light are discussed. Meteorites are protected from oxidation by a coating of vinyl chloride or in an atmosphere of dry nitrogen [M.A. 6-399]. L. J. S.

Rock-forming Minerals and Petrology.

EMMONS (R. C.). *Plagioclase determination by the modified universal stage*. Amer. Min., 1934, vol. 19, pp. 237-259, 14 figs.

A description is given of the application of the author's modified universal stage [M.A. 4-209] to the determination of plagioclases and their twins. The procedure is described and the constructions required are explained by means of diagrams and an example. The author remarks that there is no established nomenclature for the axes of the universal stage and proposes a new one, making four. The simplification and speeding up of determinations resulting from the addition of the new inner E.-W. axis is considerable. A footnote records two adjacent lamellae in a twin of plagioclase in which values of α differ by 0.000 and of β by 0.0011, indicating a compositional difference of 2-3 %. A

W. C. S.

HO (T. L.). *A rapid method for the determination of plagioclase by the Fedorov universal stage*. Amer. Min., 1935, vol. 20, pp. 790-793, 3 figs.

Suggests methods by which the procedure proposed by Emmons for the modified universal stage with five axes of rotation can be used with the Fedorov stage with only four axes of rotation. The author recommends a short cut to the usual method for finding the position of a principal axis in an optical symmetry plane. It depends on observation of the interference colours as the section is rotated [but it is not so reliable as the classical method, and may eventually entail loss of time]. He also describes a method for determining the position of a cleavage or plane of association after the position of orientation of the section is obtained, but it depends on a graphical solution. Direct measurements are not possible without the new, inner E.-W. axis, added by Emmons. An example is given of the working out of a felspar twin by the Emmons procedure.

W. C. S.

DOLAR-MANTUANI (L.). *Plagioklas aus der Leiserschlucht bei Spittal a. d. Drau*. Zeits. Krist., 1936, vol. 94, pp. 313-316, 2 figs.

Tables are given showing the position of cleavage and twin-planes, the optic axial angle, &c., for plagioclases containing various proportions of anorthite.

H. H.

ER (C. T.). *The Tertiary igneous rocks of the Pakokku district and the Salingyi township of the Lower Chindwin district, Burma, with special reference to the determination of the felspars by the Fedoroff method.* Mem. Geol. Soc. India, 1936, vol. 48, pt. 2, pp. i-xii, 221-296, i-xxvi, 11 pls., 30 text-figs. Price Rs. 4-12 = 8s.

An account of the geology of the area is followed by long descriptions of individual rock slices. Five chemical analyses are given of olivine-gabbro, gabbro, hornblende-granulite, a quartz-oligoclase-bearing hybrid and basalt. Determinations of felspars are plotted on a Reinhard diagram, and twin-laws and zoning of the felspars are discussed.

L. J. S.

EWSKA-CHLIPALSKA (Eugenia). *O składzie chemicznym pewnych adularów—Sur la composition chimique de quelques adulaires.* Arch. Min. Tow. Nauk. Warszaw. (Arch. Min. Soc. Sci. Varsovie), 1937, vol. 13, pp. 20-27 (Polish), pp. 28-31 (French résumé).

Analyses of adularia from I St. Gotthard, II Bourg d'Oisans, Dauphiné, II Krimml, Salzburg, give for $R_2O:Al_2O_3:SiO_2$ the ratios 0.9711:0.903, 0.9973:1:6.1294, and 0.9744:1:6.1164 respectively. [Cf. 5-217.]

SiO_2	Al_2O_3	Fe_2O_3	BaO	CaO	Na_2O	K_2O	H_2O-	Total.	Sp. gr.
64.70	18.66	0.20	0.78	0.19	1.74	13.44	0.09	99.80	2.5715
64.87	18.51	0.23	0.92	0.23	1.99	13.08	0.07	99.90	2.572
64.80	18.40	0.32	0.68	0.26	1.46	13.70	0.09	99.71	2.568

L. J. S.

ENBACH (E.). *Die Dichte des reinen Kalifeldspatanteils im Mikroklin und dessen allgemeine chemische Zusammensetzung.* Zentr. Min., Abt. A, 1936, pp. 231-239, 1 fig.

Analyses of microcline and microcline-perthite are calculated to mole-ratios Mi:Ab:An and plotted on a triangle. The average composition of microclines from pegmatites is Mi 75.30, Ab 23.13, An 1.57 mol. %, calculated density of Mi is 2.559. [Cf. M.A. 6-292.] L. J. S.

OBABA (Karl) & ENGELS (Agnes). *Der Einfluß der Kalifeldspatkomponente auf die Optik der Plagioklase. II. Chemismus und Optik der Endglieder Albit, Anorthit und Mikroklin.* Zentr. Min., Abt. A, 1937, pp. 103-116, 4 figs. [Cf. M.A. 6-210.]

— — III. *Die optische Orientierung kalifeldspathhaltiger Plagioklase.* Ibid., pp. 129-149, 2 figs.

Some chemical and optical data (orientation, but not refractive indices) are given for the following felspars. I, albite from Rischuna, Switzerland

land; II, anorthite, Monte Somma, Vesuvius; III, microcline, Silver Leaf mine, Manitoba; IV, albite-oligoclase, Monteagle, Ontario; V, oligoclase, Monteagle, Ontario [M.A. 5-439]; VI, andesine, Nishishiadamura, Shinano, Japan; VII, labradorite, Tamatave, Madagascar; VIII, bytownite, Crystal Bay, Minnesota. No connexion can be traced between the potash content and the migration curves of the ellipsoidal axes α , β , γ .

	SiO_2 .	Al_2O_3 .	Fe_2O_3 .	MgO .	CaO .	Na_2O .	K_2O .	ign.	Total.	Sp. g.
I.	67.45	19.68	0.40	0.13	0.26	11.20	0.32	0.35	99.79	2.62
II.	43.32	35.89	0.21	0.66	19.32	0.49	0.30	—	100.19	2.76
III.	64.03	19.12	0.37	—	0.15	2.24	14.05	0.08	100.04	2.57
IV.	65.94	20.47	0.58	—	1.54	9.91	1.09	0.25	99.78	2.62
V.	63.85	22.01	0.61	—	2.90	9.05	1.08	0.32	99.82	2.63
VI.	58.95	24.79	0.77	0.74	7.06	6.49	0.99	—	99.79	2.66
VII.	54.61	28.68	0.26	—	10.90	4.66	0.76	—	100.02	2.69
VIII.	49.46	31.73	0.49	—	14.48	3.41	0.34	0.18	100.09	2.70

VII, also TiO_2 0.15.

L. J. S.

REINHARD (M.) & BÄCHLIN (R.). *Über die gitterartige Verzwilligung beim Mikroklin.* Schweiz. Min. Petr. Mitt., 1936, vol. 16, pp. 215-225, 8 figs.

The poles of the twin lamellae and the optical directions in microclin in a gneiss from Monte Tamaro, Tessin, were determined on the universal stage and plotted. The poles of the pericline lamellae extend from (100) to (15.0.1) in the zone [010]. If the crystal is monoclinic, rather than triclinic, then the twin-axes of the pericline and albite twins coincide and the cross-hatching would be due to intercrossing albite twinning. The measurements obtained cannot decide this. [M.A. 5-217.]

L. J. S.

[SOUSTOV (N. I.)] Соустов (Н. И.). О щелочном полевом шпиле в окрестностях Христиании в южной Норвегии.—SOUSTOV (N. I.). *On the alkaline feldspar from the vicinity of Oslo in south Norway*. Труд. Петр. Инст. Акад. Наук СССР (Trav. Inst. Pétrogr. Acad. Sci. U.R.S.S.), 1936, no. 7-8, pp. 25-29. (Russian with English summary.)

A microcline-anorthoclase from a nordmarkite near Grorud, Oslo district, gave on analysis (average of two analyses): SiO_2 65.90, Al_2O_3 19.45, Fe_2O_3 1.03, MgO nil, CaO 0.61, Na_2O 7.12, K_2O 6.20, $\text{H}_2\text{O} + 0.19$ $\text{H}_2\text{O} - 0.03 = 100.53$. Sp. gr. 2.591, α 1.5261, γ 1.5314, 2V (three grains) 80° , 84° , 85° , negative.

S. I. T.

БОЛДЫРЕВ (А. К.).] Болдырев (А. К.). О морфологии, генезисе и классификации перититов и других полевошпатовых сростаний в связи с изучением Тигирецких берилловых пегматитов.—BOLDYREV A. K.). Über Morphologie, Genesis und Klassifikation der Perthite und anderer Feldspatverwachsungen in Zusammenhang mit der Untersuchung der Beryllpegmatite der Tigireklagerstätte im Altaigebirge. Груд. Центр. Научно-Писс. Геол.-Разв. Инст. (Trans. Central Geol. Prospl. Inst. U.S.S.R.), 1934, no. 12, 47 pp., 7 pls., 3 text-figs. (Russian with German summary.)

In Tigiretzk, Altai Mts., the pegmatites are composed of albite, fine-line-microperthite, muscovite, quartz, and a bluish-green beryl (als 1·5 × 0·35 m.). The classification of perthites proposed by Andersen (1928) and H. L. Alling (1932) is amplified and the genetic principles strongly emphasized. The amount of plagioclase in perthites of granite is 7 to 21 %, and from pegmatite 24 to 29 %.

S. I. T.

FER (J. W.) & THIEL (G. A.). The occurrence of fine grained authigenic feldspar in shales and silts. Amer. Min., 1937, vol. 22, pp. 842–46.

Chemical analysis of Ordovician shale from Minneapolis, Minnesota, gives the largest amounts of material as grains 1/16 and 1/512 mm. in size. The former consists mainly of quartz, and the latter is shown by optical and X-ray analysis to be mainly orthoclase. L. J. S.

MART (Duncan, Jr.). An occurrence of detrital authigenic feldspar. Amer. Min., 1937, vol. 22, pp. 1000–1003, 1 fig.

Grains of microcline isolated from glacial sand near Mt. Morris, Michigan, are surrounded by a border, probably of potash-felspar, with a different character and orientation different from those of the central part. L. J. S.

BUT (Cornelius S., Jr.). X-ray determination of the silica minerals in submicroscopic intergrowths. Amer. Min., 1936, vol. 21, pp. 727–30, 4 figs.

Quartz, tridymite, cristobalite, and perthite can be distinguished by X-ray patterns and intensity curves. Spherulites and the groundmass of rhyolite from New Mexico so examined show the presence of cristobalite and perthite. L. J. S.

[BELYANKIN (D. S.) & PETROV (V. P.)] Белянкин (Д. С.) и Петров (В. П.). О кристобалите в горных породах Кавказа и Закавказья — BELIANKIN (D. S.) & PETROV (V. P.). *Chrystobalite in the rocks of Caucasus and Transcaucasia.* Изв. Акад. Наук СССР. Отд. Мат. Ест. Наук, Сер. Геол. (Bull. Acad. Sci. U.R.S.S., Géol. Sci. Math. Nat., Sér. Géol.), 1936, pp. 303–319, 2 text-figs. (Russian with English summary.)

A review is given of the 23 occurrences of cristobalite in the Tertiary and Quaternary basalt, andesite, trachyandesite, trachyte, and dacite of the main Caucasian chain and Transcaucasia. S. I. T.

Fox (Cyril S.). *The solubility of quartz.* Rec. Geol. Survey India, 1931, vol. 69, pp. 423–425, 1 pl.

Quartzite pebbles in conglomerate of the coal measures of Jharia show signs of corrosion, attributed to solution by surface waters.

L. J. S.

COOKE (H. C.). *An unusual hypersthene from Lake Athabaska, Saskatchewan.* Univ. Toronto Studies, Geol. Ser., 1937, no. 40, pp. 67–68.

A brecciated norite partly replaced by nickeliferous pyrrhotine, consists of light grey hypersthene and felspar ($\text{Ab}_{55-60}\text{An}_{45-40}$) with very little accessory biotite and garnet. The hypersthene is colourless in thin section, n 1.70–1.71, birefringence 0.015–0.016, $2V$ 61–63°, negative. Analysis by R. J. C. Fabry gave SiO_2 48.91, TiO_2 0.83, Al_2O_3 3.50, Fe_2O_3 0.30, FeO 24.16, MnO 0.16, MgO 19.59, CaO 0.59, Na_2O 0.05, K_2O 0.12, $\text{H}_2\text{O} + 0.99 = 99.16$. L. J. S.

VERHOOGEN (Jean). *A monoclinic "hypersthene" from the Cascade lava.* Amer. Journ. Sci., 1937, ser. 5, vol. 33, pp. 63–69, 2 figs.

Optical data are given for pyroxenes in andesites and dacites from the Cascade Range in Washington and Oregon. These include hypersthene (α 1.676, β 1.687, γ 1.695, $2V$ 75–85°, negative), diopsidic augite (α 1.69–1.695, β 1.695–1.715, γ 1.715–1.725, $2V$ 60°, $\gamma:c = 43-51^\circ$), pigeonite ($2V$ 40–45°, $\gamma:c = 35^\circ$), and a monoclinic 'hypersthene'. The last resembles hypersthene in appearance and pleochroism, but in the zoned crystals gives extinction-angles ranging from 4° in the core to 18° in the rim; $2V$ 65–85°, negative. From a rock analysis the composition of this mineral is deduced as SiO_2 57.0, Al_2O_3 1.2, FeO 16.8, MgO 20.4, CaO 3.2. The optical data differ from those of clinohypersthene [M.A. 6–211]. L. J. S.

HERT (Róbert). *Kristálytani megfigyelések egy bőrzsönyi andezittufa éhány ásványán*.—REICHERT (Robert). *Kristallographische Beobachtungen an einigen Tuffmineralien aus dem Börzsönyer-Gebirge*. Földtani Közlöny, Budapest, 1935, vol. 65, pp. 342–349, 4 figs. (German with Hungarian summary.)

Augite crystals from andesite-tuff show the forms $a\ b\ m\ s\ c\ p\ o\ z$ [a's letters]; $\gamma:c = 38\text{--}40^\circ$, $\alpha 1\cdot678$, $\gamma 1\cdot702$. Common green hornblende shows the forms $m\ b\ c\ d\ u\ a, z$ (201), and k (111); $\gamma:c = 16^\circ$, $\alpha 1\cdot680$, $\gamma 1\cdot680$. Garnet crystals show zonal structure but are perfectly euhedral; they enclose plagioclase, magnetite, hypersthene, biotite, and ilmenite.

V. Z.

KELLY (Norman). *A petrological study of the Portrush sill and its veins*. Proc. Roy. Irish Acad., 1937, vol. 43, sect. B, no. 9, pp. 95–134, 3 pls., 10 text-figs.

A thick sill of olivine-dolerite in Co. Antrim has a roof of hornfelsed dolerite and is penetrated by veins of hornfels, porphyrite, &c. Five types of clinopyroxene are distinguished: augite (2V about 58° , $\gamma:c = 40^\circ$, anal. I); hypersthene-augite (2V about 60° , $\gamma:c = 50^\circ$); pigeonite (2V 38° , $\gamma:c = 45^\circ$, $\alpha 0\cdot022$, anal. II); hypersthene (2V $52\text{--}54^\circ$, $\gamma-\alpha 0\cdot015$); and diopside (2V $58\text{--}60^\circ$, $\gamma:c = 42^\circ$). Anal. III, marginal type of olivine-dolerite.

SiO_2	TiO_2	Al_2O_3	Fe_2O_3	FeO	MnO	MgO	CaO	Na_2O	K_2O	$\text{H}_2\text{O}+$	$\text{H}_2\text{O}-$
41·05	0·50	5·23	0·90	7·35	0·25	14·18	19·10	0·39	0·07	0·50	0·50
0·15	2·40	1·95	2·15	17·22	0·65	13·08	11·02	0·38	0·12	0·60	0·40
6·51	0·84	15·60	0·99	9·14	0·16	9·05	11·77	1·89	0·72	1·79	0·75
and II, also P_2O_5 trace.											
III, also Cr_2O_3 0·02, V_2O_3 0·02, NiO 0·07, CuO 0·15,											
0·02, BaO 0·04, CO_2 0·07, P_2O_5 0·52, F 0·04, Cl 0·04, S 0·15.											

L. J. S.

WILLIAMS (W. A.). *Note on a pegmatitic hornblende from the Carsphairn complex [Kirkcudbrightshire]*. Geol. Mag. London, 1937, vol. 74, pp. 359–361.

In a pegmatite, consisting of quartz, hornblende, and sphene, forms large crystals of late residuum crystallization in the hornblende hybrids. The darkish-black crystals of hornblende have $\alpha 1\cdot659$, $\beta 1\cdot669$, $\gamma 1\cdot680$, $\delta 1\cdot69^\circ$, $\gamma:c = 24^\circ$. Analysis gave SiO_2 50·18, TiO_2 1·58, Al_2O_3 4·27, FeO 2·78, MnO 0·30, MgO 11·75, CaO 11·84, Na_2O 0·95, K_2O 0·51, $\text{H}_2\text{O}+$ 1·93, $\text{H}_2\text{O}-$ 0·20 = 100·47; formula $(\text{OH})_{1\cdot9} (\text{Na}, \text{Ca}, \text{K})_{2\cdot2} (\text{Mg}, \text{Fe}^{\prime\prime}, \text{Fe}^{\prime\prime\prime}, \text{Ti}, \text{Mn}, \text{Al})_5 [(\text{Si}, \text{Al})\text{O}_{22}]$.

L. J. S.

ESKOLA (Pentti) & KERVINEN (Tauno). *A paragenesis of gedrite and cummingtonite from Isopää in Kolvola, Finland.* Bull. Comm. Géol. Finl. 1936, no. 115 (= Compt. Rend. Soc. Géol. Finl. no. 9), pp. 475-487, 4 figs.

Garnet-amphibolite contains in a groundmass of plagioclase (Ab_3An_1) and quartz: I, greyish-green gedrite (sp. gr. 3.371, $\alpha' 1.674$, $\gamma' 1.697$); II, colourless cummingtonite (sp. gr. 3.307, $\alpha' 1.661$, $\gamma' 1.681$); and III, almandine crystals up to 6 cm. across ($n 1.810$).

	SiO_2	TiO_2	Al_2O_3	Fe_2O_3	FeO	MnO	MgO	CaO	Na_2O	K_2O	H_2O
I.	43.70	0.55	10.88	3.52	26.53	0.24	11.48	0.54	1.24	0.15	1.2
II.	50.70	0.31	1.72	3.11	26.63	0.19	14.36	0.87	0.60	0.15	1.4
III.	37.31	0.72	20.00	—	34.56	1.00	3.89	2.48	—	—	—

L. J. S.

WAYLAND (Russell G.). *Cummingtonite from the Black Hills, South Dakota.* Amer. Min., 1936, vol. 21, pp. 607-610.

Brownish or greenish-grey, bladed or fibrous cummingtonite occur in metamorphic rocks in the Homestake mine and at other localities near Lead. In fourteen specimens FeO ranges from 21.54 to 34.45 %, with corresponding variations in the physical data. Sp. gr. 3.19-3.40, $\alpha 1.653-1.672$, $\beta 1.670-1.694$, $\gamma 1.682-1.704$, $\gamma - \alpha 0.025-0.040$, $\gamma : c = 17-24^\circ$, $2V_a 70-98^\circ$ (those low in FeO are positive and with higher extinction-angle). Dimensions of the unit cell, containing two molecule $(\text{OH})_2(\text{Fe},\text{Mg})_7(\text{Si}_4\text{O}_{11})_2$, $a 9.50-9.55$, $b 18.18-18.23$, $c 5.33-5.35 \text{ \AA}$. b and c seem to increase slightly with iron. Space-group $C_{\bar{2}h}^1$.

L. J. S.

RONDOLINO (Rinaldo). *Sopra alcuni anfiboli manganesiferi di Praborno (San Marcel—Valle d'Aosta).* Periodico Min. Roma, 1936, vol. 7, pp. 109-121. [Cf. M.A. 6-85.]

The following manganese-bearing amphiboles are associated with braunite, spessartine, violane, &c. I, Colophony-brown fibres and laminae, of the richterite type; metasilicate formula $\text{H}_8\text{R''}_{13}\text{Si}_{17}\text{O}_{51}$ with $\text{R}'_2\text{R''''}_2\text{SiO}_6$ in solid solution. II, Rosy fibres with violet reflections, also of the richterite type; $\text{H}_6\text{R''}_{12}\text{Si}_{15}\text{O}_{45}$ with $\text{R}'_2\text{R''''}_2\text{SiO}_6$. III, Straw-yellow flexible fibres, of the edenite type.

	SiO_2	Al_2O_3	Fe_2O_3	MnO	MgO	CaO	Na_2O	K_2O	$\text{H}_2\text{O}+$	$\text{H}_2\text{O}-$
I.	55.09	2.35	4.54	4.70	17.88	7.99	3.37	0.32	3.63	0.36
II.	55.23	6.06	trace	0.85	20.80	9.49	2.72	0.94	2.80	0.25
III.	48.28	8.55	1.58	6.76	18.29	9.46	3.94	0.26	2.44	0.32

Total.	Sp. gr.	α .	β .	γ .	$\gamma:c$.	2V.
100.23	3.00	1.610	1.627	1.639	22°	79° 12'
99.14	2.95	1.602	1.618	1.630	25	80 48
99.88	3.05	1.627	1.636	1.641	20	70 4

L. J. S.

ДВНИК (Н. И.)] НАКОВНИК (Н. И.). Синие амфиболы магнетитовых кварцитов Курской магнитной аномалии.—NAKOWNIK (N. J.). *blaue Amphibolen der Eisenquarze aus der Gegend der Magnet-anomalie von Kursk*. Зап. Всеросс. Мин. Общ. (Mém. Soc. Russe Min.), 1934, ser. 2, vol. 63, pp. 94–108, 7 figs. (Russian with German summary.)

zonal amphiboles from the magnetite-quartzite of the Kursk area are subdivided into five types in order of the increasing intensity of pleochroism (α yellow, β violet, γ blue). Types 1 and 2 are close to olivine-tremolite, with optic axial plane parallel to (010) and 2V ranging from 60° to 0° (between types 2 and 3). Beginning with type 3 the dispersion of the optic axes and bisectrices is gradually lessening and the optic axial plane is perpendicular to (010) with the exception of the 4th type in which the optic axial plane for violet is parallel to (010) and for red perpendicular to (010). The increase of 2V is marked. The refractive index varies from α 1.636, γ 1.649 (type 1) to α 1.677, γ 1.685 (type 5). It is assumed that the blue amphiboles are rich in soda, but no analyses are given.

S. I. T.

F (John E.). *Hastingsite in therelite from the Crazy Mountains, Montana*. Amer. Min., 1937, vol. 22, pp. 742–744.

Hastingsite as grains or sometimes in parallel growth with augite and iron-augite has α (dull yellow) 1.639, β (liver-brown) 1.658, γ (brownish-brown) 1.660, $\gamma:c = 38^\circ$, 2V 38°, negative, $r < v$ strong. Analysis by F. A. Gonyer gave SiO₂ 48.51, TiO₂ 1.32, Al₂O₃ 6.60, FeO 4.09, FeO 9.48, MnO 0.19, MgO 14.79, CaO 5.60, Na₂O 6.01, H₂O + 1.47 = 100.26, sp. gr. 3.23; (Na,K)₂Ca(Mg,Fe'',Fe'')₂(Al,Si)₈(OH)₂O₂₂. A chemical analysis is given of the rock.

L. J. S.

LEY (J. E.). *Riebeckite in quartz veins from the Michipicoten district, Ontario*. Amer. Min., 1937, vol. 22, pp. 1099–1103.

Clusters of fibrous riebeckite occur in quartz veins, with gold and other minerals, traversing granite and the adjoining basic volcanic rocks. α 1.699; $\alpha:c = 5\frac{1}{2}^\circ$ (600 $\mu\mu$), 7° (589 $\mu\mu$), 9° (550 $\mu\mu$). The mineral replaces mica, epidote, &c., and is of hydrothermal origin. L. J. S.

BOWEN (Norman L.). *A note on aenigmatite.* Amer. Min., 1937, vol. 22, pp. 139-140.

Phenocrysts in a pantelleritic lava from Lake Naivasha, Kenya, gave α 1.81, β 1.82, γ 1.88 (the birefringence being much higher than previously recorded for aenigmatite), extinction 38° ; strong pleochroism α yellow-brown, β brown, γ deep brown to black. L. J. S.

MEHMEL (M.). *Ab- und Umbau am Biotit.* Chemie der Erde, 1936, vol. 11, pp. 307-332, 8 figs.

— *Thermischer Umbau am Biotit.* Fortschr. Min. Krist. Petr., 1936, vol. 22, pp. xlvii-xlviii, 1 fig.

The biotite used in the experiments was from Ketilä, Impilahti, Finland; analysis gave SiO_2 35.96, TiO_2 2.55, Al_2O_3 12.50, Fe_2O_3 7.8, FeO 15.12, MnO 1.02, MgO 9.29, Na_2O 1.96, K_2O 8.56, H_2O 4.96 = 99.73; β and γ 1.650, 2V near 0° . Strong acid (1/1 H_2SO_4) passed over the small flakes for many hours extracted all the bases together with some silica, leaving white flakes of amorphous silica (n 1.420). Weak acid acts in the same direction but much slower. Alkali produces a red iron coating on the flakes. Heated at $450-700^\circ$ the biotite acquires yellow colour and n 1.823, 2E 82° . It is suggested that bleached biotite (bauerite) in sedimentary rocks has been confused with muscovite and clay minerals.

L. J. S.

HESS (Frank L.) & STEVENS (Rollin E.). *A rare-alkali biotite from King Mountain, North Carolina.* Amer. Min., 1937, vol. 22, pp. 1040-1044, 1 fig.

Biotite from the border of a spodumene-pegmatite in mica-schist gave α 1.555, β 1.589, γ 1.590, 2V $0-5^\circ$; SiO_2 42.02, TiO_2 1.35, Al_2O_3 18.75, Fe_2O_3 0.66, FeO 8.29, MnO 0.27, MgO 9.55, CaO 0.93, Li_2O 1.20, Na_2O 0.73, K_2O 8.54, Rb_2O 1.85, Cs_2O 0.47, $\text{H}_2\text{O}+$ 2.44, $\text{H}_2\text{O}-$ 0.16, F 4.34 total (less O for F) 99.72. Dark micas from the border of pegmatite from several other localities were examined spectroscopically for rare alkalis, and a redetermination of these in the 'caesium biotite' from Ti Mtn., South Dakota, gave Li_2O 0.65, Rb_2O 1.48, Cs_2O 1.12 [instead of 3.14, M.A. 5-192].

L. J. S.

GALLITELLI (Paolo). *Le miche del granito di Baveno.* Periodico Min. Roma, 1936, vol. 7, pp. 61-76, 4 figs.

The micas of the Baveno granite include: siderophyllite (I, green from a fayalitic nodule; II, greenish-black from a basic patch); lepto-

zinnwaldite and protolithionite (zoned crystals with variable data); and muscovite (III, green resembling talc).

SiO ₂ .	TiO ₂ .	Al ₂ O ₃ .	Fe ₂ O ₃ .	FeO.	MnO.	MgO.	CaO.	Na ₂ O.	K ₂ O.
32.09	nil	13.33	17.75	19.88	0.31	2.19	1.15	1.50	6.30
30.49	1.17	16.40	14.01	18.00	0.69	2.40	1.10	4.88	5.65
47.64	nil	28.60	2.06	3.15	0.08	nil	trace	9.64	2.04
H ₂ O.	H ₂ O+.	H ₂ O-.	F.	Sp.gr.	α .	β .	γ .		2E.
nil	4.68	0.60	0.30	3.12	1.612—	1.667—	—		3—6°
					1.614	1.669			
nil	3.79	0.93	0.61	3.06	—	—	—	—	—
0.91	4.39	—	2.80	—	—	1.582—	1.585—	36—39°	
						1.584	1.587		

L. J. S.

MATARI (Mititaka). *A study on the influence of original composition upon the micas in the metamorphic rocks of the Hsian shui ssu series, in the vicinity of Mt. Ta ho shang, south Manchuria.* Mem. Ryojun Coll. Engineering, 1935, vol. 8, no. 4, pp. 67—84, 1 pl., 5 text-figs. Technical analyses are given of the schists and optical data for the pyroxite and biotite.

L. J. S.

EN (N. L.) & ELLESTAD (R. B.). *Nepheline contrasts.* Amer. Min., 1936, vol. 21, pp. 363—368, 1 fig.

Nephocrysts of fresh and glassy nepheline from a phonolite lava flow 10 miles north of Nairobi, Kenya, gave analysis I; ω 1.535, ϵ 1.531. Nephocrysts from a lava flow of melilite-leucite-nephelinite near Burera, north of Lake Kivu, Belgian Congo, gave II; ω 1.544, ϵ 1.540. Composition is expressed as solid solutions of Na₈Al₈Si₈O₃₂(Ne), Ca₈Si₈O₃₂(Kp, kaliophilite), Na₄Al₄Si₁₂O₃₂(Ab), and Ca₄Al₄Si₈O₃₂(An), of which molecules conforms with the nepheline formula n Al _{n} (Na,K, $\frac{1}{2}$ Ca) _{n} O₃₂ of Bannister and Hey [Min. Mag. 22—580]. I (61.8, Kp 12.9, An 4.5, Ab 20.8) is the richest in silica, and II (77.0, Kp 37.6, An 4.4, Ab 1.0) the richest in potash, of any nephelines previously analysed. Analyses are also given of the rocks, showing that phonolite from the eastern Rift is richer in soda, while the nephelinite from the western Rift is richer in potash, corresponding with the composition of the contained nepheline.

SiO ₂ .	Al ₂ O ₃ .	Fe ₂ O ₃ .	MgO.	CaO.	Na ₂ O.	K ₂ O.	H ₂ O+.	H ₂ O-.	Total.
46.41	31.07	0.78	0.11	0.87	15.67	3.81	0.97	0.17	99.86
40.74	33.39	0.83	0.25	0.91	12.53	11.13	0.23	0.06	100.18

II, also TiO₂ 0.11, MnO trace.

L. J. S.

BOWEN (N. L.) & ELLESTAD (R. B.). *Leucite and pseudoleucite.* Amer. Min., 1937, vol. 22, pp. 409–415, 2 figs.

Porphyritic crystals of leucite in leucitophyre show white rims consisting of a mixture of orthoclase and nepheline; and leucite in the groundmass of leucite-theralite is often altered to the same minerals. Both of these leucites contain only 1–1·5 % Na₂O, and their alteration to nepheline indicates that soda was introduced by reaction with the magma. Analyses of these rocks, from Lake Kivu, Belgian Congo, are given.

L. J. S.

KUNITZ (Wilhelm). *Beitrag zur Kenntnis der magmatischen Assoziationen III. Die Rolle des Titans und Zirkoniums in den gestein bildenden Silikaten.* Neues Jahrb. Min., Abt. A, 1936, Beil.-Bd. 7, pp. 385–466, 11 figs.

The composition of rock-forming silicates containing titanium and zirconium is discussed in detail. Ti^{IV} replaces Si in titangarnets, titan-hornblendes (aenigmatite), titanmicas, titantourmaline, and titan-olivine; Ti^{III} replaces Al in titanaugites; and Ca-Ti may be replaced by rare-earths, Al, Fe^{II} in sphene (grothite and keilhauite). Replacements Zr-Hf are found in zircon (malacon, cyrtolite) and eudialyte-Zr-Ti in the astrophyllite group; and Ti(Zr)-Nb(Ta) in the lavenei group. A bibliography of 222 items is added.

L. J. S.

TSUBOI (Seitarō). *Petrological notes (11)–(18).* Japanese Journ. Geogr., 1936, vol. 13, pp. 333–337, 1 fig. [Cf. M.A. 6–210.]

Pumpellyite from Asahine, Titibu, Saitama, is a green fibrous mineral filling veins in basic igneous rocks; it is monoclinic with elongation $\parallel b$ -axis and twinned on (001); positive, $\beta = b$, $\alpha:a = 11\text{--}12^\circ$; pleochroism α and γ almost colourless, β pale green; at 600° C. 1·29 % H₂O is lost without change in the crystal-structure, and at 800° C. 6·51 % H₂O is lost with change in the crystal-structure; analysis I. [Cf. Min. Mag. 24, 529.] Piedmontite from Ikadadu, Ehime, as crystals in sericite-piedmontite-quartz-schist; anal. II after deducting 2·77 % quartz. Biotite in uncontaminated biotite-granites from III Iwakura, Yamaguti IV Tutiyane, Yamaguti, and V Sanage-mura, Aiti. Hornblende in amphibolite from Ipponmatu, Ehime; $\gamma:c = 18^\circ$, anal. VI. Iso-orthoclase in hornfels from Sakuma-mura, Siduoka; 2V_y ranges from 83° (positive iso-orthoclase) to 100° (negative, orthoclase). Cordierite in the same rock shows a still wider range in 2V from near 0° over γ (positive) to

10° over α (negative); in one crystal $2V$ about 80° , negative, 5; and in another $2V$ near 0° , positive, β 1.560. All analyses by Iaka.

SiO ₂ .	TiO ₂ .	Al ₂ O ₃ .	Fe ₂ O ₃ .	FeO.	MnO.	MgO.	CaO.	Na ₂ O.	K ₂ O.
37.53	0.14	25.39	1.74	2.95	0.13	2.65	22.56	0.58	nil
38.40	0.05	22.75	10.55	nil	3.14	0.30	22.70	0.23	trace
33.68	2.28	17.18	2.42	25.60	1.66	4.46	0.80	0.48	7.37
34.60	3.33	14.28	2.10	26.80	0.50	4.98	0.60	0.41	8.53
33.86	2.74	14.93	3.02	26.02	0.84	6.03	0.39	0.50	8.31
46.06	0.37	13.80	2.26	6.47	0.18	14.78	10.16	3.12	0.39
IP ₂ O ₅ .	H ₂ O+.	H ₂ O-.	Total.	Sp. gr.	α .	β .	γ .	2V.	
trace	6.37	0.09	100.13	3.2	1.686	1.690	1.698	36-40°	
trace	1.68	0.20	100.00	—	1.730	1.744—	1.765	74-90	
						1.747			
0.23	3.50	0.15	99.81	—	—	—	1.673	—	
0.08	3.30	0.15	99.66	—	—	—	1.670	—	
0.08	3.10	0.43	100.25	—	—	—	1.672	—	
trace	1.78	0.11	99.48	—	1.637	—	1.657	85	

L. J. S.

ERUP (Niels-Henr.). *Korund, Högbonit, Staurolith und Skapolith in den Anorthositgabbros des Bergengebietes.* (Vorläufige Mitteilung.) Bergens Museums Årbok, for 1936, 1937, no. 8, 11 pp., 3 figs.
These minerals were recognized in thin sections of the rock.

L. J. S.

EV (Esper S.), IRVING (John), GONYER (F. A.), & LARSEN (Esper S., 3rd). *Petrographic results of a study of the minerals from the Tertiary volcanic rocks of the San Juan region, Colorado.* Amer. Min., 1936, vol. 21, pp. 679-701, 8 figs.; 1937, vol. 22, pp. 889-905, 5 figs.

The occurrence of different silica minerals in the various rocks (silites, quartz-latites, with some andesites and basalts) is described, the conditions of their formation are discussed. Tridymite forms during rapid crystallization in the porous parts of the groundmass; and here rocks predominates over quartz, giving tridymite-rhyolite and tridymite-latite. Cristobalite is usually found in gas cavities. Chemical analyses and optical data are given for pyroxenes and olivine, and the conditions of formation of pigeonite are discussed. I (also F trace), diopside phenocrysts in quartz-latite from Treasure Mtn. II, diopside phenocrysts in andesite-latite from Los Pinos. III, diopside phenocrysts in pyroxene-andesite from Sheep Mtn. IV, hypersthene phenocrysts in the same rock as III. V, olivine in basalt from Buffalo Buttes. The second part of the paper deals with the amphiboles and biotite,

giving several chemical analyses and other data. These minerals are often resorbed and replaced by iron oxides. The conditions that determine the crystallization of pyroxene or hornblende are discussed.

	SiO ₂ .	TiO ₂ .	Al ₂ O ₃ .	Fe ₂ O ₃ .	FeO.	MnO.	MgO.	CaO.	Na ₂ O.	K
I.	51.05	0.36	1.80	2.03	6.56	1.22	13.82	22.06	0.38	0
II.	51.45	0.36	1.50	3.15	6.45	0.53	14.11	21.69	0.32	1
III.	50.74	0.95	2.98	2.37	10.04	0.17	14.24	17.88	—	0.67
IV.	51.58	0.45	1.70	3.15	18.11	0.39	22.01	1.82	—	1
V.	39.31	0.06	1.68	—	19.84	0.17	37.74	0.87	—	1
	H ₂ O+.	H ₂ O—.	P ₂ O ₅ .	Total.	Sp. gr.	α.	β.	γ.	2V.	γ.
I.	0.17	0.04	0.46	100.03	3.370	1.684	1.691	1.712	60°	1
II.	0.16	nil	0.48	100.20	3.37	1.686	1.693	1.713	64	1
III.	0.17	0.03	—	100.24	3.405	1.695	1.701	1.719	55	1
IV.	0.40	—	—	99.61	3.481	1.702	1.707	1.712	—	1
V.	—	—	—	99.67	—	1.675	1.694	1.712	90	1

L. J. S.

WELLS (Roger C.). *Analyses of rocks and minerals from the laboratory of the United States Geological Survey 1914–36*. Bull. U.S. Geol. Surv., 1937, no. 878, x+134 pp. Price 15 cents.

This supplements Bulletin 591, collecting together 1533 detailed chemical analyses made during the period 1914–36. Some of them have not been published previously. Analyses of igneous and crystalline rocks, sandstone and chert, carbonate rocks, slates and shales, clays and soils, ores and gangue minerals are each grouped under States. Analyses (515) of minerals are classified chemically under elements, sulphides, &c. A table gives the relative abundance ('clarke's') of the chemical elements according to the revision of A. E. Fersman, 1933 [M.A. 5–433].

L. J. S.

[ZAVARITZKY (A. N.)] Заваричский (А. Н.). Об одной важной петрохимической закономерности.—ZAVARITSKY (A. N.). An important petrochemical peculiarity of igneous rocks. Зап. Ленинград. Геол. Инст. (Ann. Inst. Mines Leningrade), 1935, vol. 9, pp. 1–11 pl. (Russian with English summary.)—

By recalculating, according to his own system of ratios, 135 analyses of igneous rocks given by Daly [M.A. 5–290] and by plotting the graphically, the author derives the following conclusions. In acid rocks the felsic portion is rich in iron oxides, in basic rocks, in magnesia and in alkali rocks, in lime.

S. I. T.

EN (Norman L.). *Recent high-temperature research on silicates and its significance in igneous geology.* Amer. Journ. Sci., 1937, ser. 5, vol. 33, pp. 1-21, 10 figs.

Study of the several systems involving SiO_2 , Al_2O_3 , FeO , CaO , MgO , K_2O shows that with fractional crystallization the residue consists of alkali-Al silicate, NaAlSiO_4 - KAlSiO_4 - SiO_2 , which is referred to as retrogeny's "residua" system'. Plots of analyses of phonolites, leucites, and rhyolites from the East African Rift Valley fall within this area. It is suggested that these rocks represent the residual liquids of crystallizing magmas.

L. J. S.

GRIEVEV (D. P.) & ISKYUL (E. V.)] Григорьев (Д. П.) и Искюль (Е. В.). Дифференциация некоторых силикатных расплавов как результат образования двух несмешивающихся жидкостей.— GRIORIEV (D. P.) & ISKÜLE (H. V.). *Differentiation of some silicate melts due to the formation of two immiscible fluids.* Изв. Акад. Наук СССР, Отд. Мат. Ест. Наук, Сер. Геол. (Bull. Acad. Sci. U.R.S.S., Cl. Sci. Math. Nat., Sér. Géol.), 1937, pp. 77-106, 14 figs. (Russian with English summary.)

Silicate melt with CaF_2 and K_2CO_3 as fluxes (anal. I) showed a distinct gravitational separation of two glassy phases. The lower layer (II) is an emulsion with minute glassy globules enclosed in a glassy groundmass (anal. III). The upper layer (anal. IV), sharply separated from the lower layer, contains a number of gas inclusions. The disperse phase (globules) of the lower layer is considered to represent the material of the upper layer, which was arrested during the gravitational separation of the two liquids. It is suggested that the liquid immiscibility is mainly determined by the presence of fluorine, and that boron and silicon may probably have the same effect. A comparison is made with a leucite from Yalguba [M.A. 6-303] and other rocks, and the question of differentiation discussed.

SiO_2 .	MgO .	CaO .	Al_2O_3 .	F.	Total.	Refr. index.	Sp. gr.
53.15	14.28	12.41	11.95	3.06	94.85	—	—
47.35	18.31	15.23	12.49	2.30	95.68	—	2.452
34.46	27.26	21.50	13.67	0.63	97.52	1.582	2.540 (calc.)
66.69	4.20	5.82	10.68	4.82	'92.91'	1.490- 1.505	2.321

S. I. T.

A.—C

[VOLAROVICH (M. P.), TOLSTOY (D. M.), & KORCHEMKIN (L. I.).
 VOLAROVIČ (M. P.), TOLSTOJ (D. M.), & KORČEMKIN (L. I.). *A study of the viscosity of molten lavas from Mount Alaghez.* Compt. Rend. (Doklady) Acad. Sci. U.R.S.S., 1936, vol. 1, pp. 333–336, 2 figs.

Viscosity measurements made with a rotating cylinder show the relations existing between the composition of the rocks and the viscosity of their melts at different temperatures. S. I. T.

[KORZHINSKY (D. S.)] Коржинский (Д. С.). Подвижность инертность компонентов при метасоматозе.—KORŽINSKIJ (D. S.). *Mobility and inertness of components in metasomatism.* Изв. Акад. Наук СССР, Отд. Мат. Ест., Сер. Геол. (Bull. Acad. Sci. U.R.S.S., Cl. Sci. Math. Nat., Sér. Géol.), 1936, pp. 35–57 (Russian), pp. 5, 60 (English).

The application of the phase rule to the rock-solution system allows the calculation of the relative mobility of the components during the process of metasomatism. It is found that the mobilities of the components are roughly proportional to their ionic radii. S. I. T.

GILLULY (James). *The water content of magmas.* Amer. Journ. Sci., 1937, ser. 5, vol. 33, pp. 430–441.

A critical review of the literature, with the suggestion that de-seated basaltic magmas may contain 4 % and granitic magmas 8 % H₂O. L. J. S.

GORANSON (Roy W.). *Silicate-water systems: the "osmotic pressure" of silicate melts.* Amer. Min., 1937, vol. 22, pp. 485–490, 1 fig.

Differential hydrostatic pressure on silicate and water would occur if the country-rock surrounding a magma were pervious to water and impervious to the silicate melt. This problem is considered theoretical. L. J. S.

WELLS (A. K.). *Petrographic nomenclature.* Geol. Mag. London, 1937, vol. 73, pp. 319–325.

A committee appointed by the British Association for the Advancement of Science to consider and report upon petrographic nomenclature and classification has put forward for criticism a proposal to establish a classification of igneous rocks into three divisions based on grain-size in place of the classification widely, but loosely, used into three divisions based on mode of occurrence. In order to illustrate the proposed scheme

details are given for the subdivision of the gabbro-dolerite-basalt, and of the granite-microgranite-rhyolite series. Different limits in-size are used in these two series, but the proposals are in practice as complicated as they may appear at first sight. W. C. S.

· (John C.). *Igneous rock names and their evaluation*. Amer. Min., 1936, vol. 21, pp. 427-441.

The haphazard naming of rocks and rock series and the re-definition of earlier names are discussed. Names derived from localities are deprecated, and many of the names are quite unnecessary. L. J. S.

GOLDS (Doris L.). *Augite-biotite-diorite of the Newry complex*. Geol. Mag. London, 1937, vol. 74, pp. 476-477.

The rock described as 'augite-biotite-diorite' (Quart. Journ. Geol. London, 1934, vol. 90, pp. 609-611), and later in discussion called 'acid sodi-potassic gabbro', 'gabbro-diorite', and 'biotite-essexite-co', is now named *garronite*, from the hill Slievegarron, Co. Down.

L. J. S.

KUPLETSKY (B. M.)] Куплетский (Б. М.). К вопросу о генезисе щелочных пород.—KUPLETSKI (B. M.). *On the genesis of alkaline rocks*. Изв. Акад. Наук СССР, Отд. Мат. Ест., Сер. Геол. (Bull. Acad. Sci. U.R.S.S., Cl. Sci. Math. Nat., Sér. Géol.), 1936, pp. 329-336 (Russian), pp. 337-339 (English).

A single theory can account for all alkali rocks. The Kola peninsula displays three genetic types: (1) associated with alkali-granites (Khibina, Lovozero tundras); (2) associated with basic rocks (Gremyakha tundra, Afrikanda); (3) a possible case of limestone-magma axis (Turya peninsula, river Kovdora) [M.A. 6-305-313].

S. I. T.

JANN (Eugène). *Sur la genèse des roches alcalines de Julianehaab (Groenland)*. Compt. Rend. Acad. Sci. Paris, 1937, vol. 204, pp. 1125-1127.

Further observations on the spot modify the old view that these rocks are formed by differentiation due to separation of the successively crystallized portions by gravity, and it is now thought that they consist of a series of intrusive rocks from essexite through nordmarkite to melsonite-granite, resulting from intrusive masses which have digested

surrounding rocks, and (b) a group of strongly metamorphic rock lujaurites, naujaites, kakortokites, foyaites and other nepheline syenites, the transformation of which occurred subsequently to the intrusion of (a). They appear to consist of an original rock of which the structure is more or less preserved, which has been chemically transformed by additions from elsewhere, but without actual fusion occurring.

C. A. S.

DIXEY (F.), SMITH (W. Campbell), and BISSET (C. B.). *The Chilwa Series of southern Nyasaland; a group of alkaline and other intrusive and extrusive rocks and associated limestones.* Bull. Geol. Surv. Nyasaland, 1937, no. 5, 82 pp., 7+8 pls., 1 text-fig. Price 5s. Abstract in Abstr. Proc. Geol. Soc. London, 1936, session 1936-37, pp. 8-10.

The rocks described, for which the term 'Chilwa Series' is suggested, occur in the vicinity of Lake Chilwa and elsewhere in southern Nyasaland, as well as in the adjacent part of Portuguese East Africa. They comprise principally the following rocks: syenite, volcanic vents infilled with carbonates and brecciated felspar-rock, a minor development of hydrothermal rocks in the vents, nepheline-syenite, and swarms of associated dikes of sölvsbergite, microfoyaite, phonolite, and nephelinit. The intrusions and eruptions represent a hitherto unrecognized phase of Karroo or early post-Karroo igneous activity. The volcanic vents range from some only a few yards across to one nearly four miles in diameter. The country-rocks are extensively altered and the vents are filled with highly felspathic agglomerates and breccias associated in five of the larger vents with great masses of crystalline calcium carbonate, and to a less extent of iron and manganese carbonates, and cut by nepheline-syenite and nephelinites. The felspathic material of the vents is remarkably rich in orthoclase of unusually high potash content. The alteration of the surrounding gneisses results in the introduction of aegirin, augite at the expense of biotite and other minerals, and the altered rocks are comparable to the fenite-tveitåsite series described by Brögger from the Fen district of Telemark, Norway, where also they are associated with masses of limestone and ijolitic intrusives. The nepheline-syenite and the minor intrusions present a great variety of rock types containing many interesting minerals, some of which remain to be identified. The amphibole imerinite is recognized in the altered gneisses, and a mineral resembling molengraaffite appears in some of the microfoyaite. Analyses of some of the rocks have been made by M. H. Hey. Attention is called to the existence of alkali and other igneous rocks mainly

oo age in other parts of Africa which are comparable with the wa Series, and comparisons are made also with other alkali rocks riated with crystalline limestones. The vents appear to have been ed at least in part by explosive action, and this may be closely coned with the presence of the carbonates. The limestones are regarded very deep-seated, possibly magmatic, origin, comparable with the matic limestones or carbonatites of Alnö in Sweden and of the Fen äct in Norway, but developed on a scale far larger than at either of e localities. Their relations to the orthoclase-rocks and to the neph- syenites are to be discussed after a more detailed examination has made of the crystalline limestones.

W. C. S.

MES (Arthur) & HARWOOD (H. F.). *The volcanic area of Bufumbira. Part II: The petrology of the volcanic field of Bufumbira, south-west Uganda, and of other parts of the Birunga field.* Mem. Geol. Surv. Uganda, 1937, no. 3, pt. 2, xiv+300 pp., 5+9 pls. Price 21s.

The first part of this memoir, dealing with the geology of the area, by A. D. Combe and W. C. Simmons, and appeared in 1933. Inten-work has since been done on the materials they collected, and this d part contains a mass of detail that cannot be condensed into an tract. The volcanic rocks described include: the ugandite-olivine- tite series, leucitite-nephelinite series and vesuvite, kivite-leucite- nite series, murambite-absarokite group, shoshonitic absarokite- kite-trachyte series, shoshonite-latite series, and the limburgite- nybasalt series. Plutonic rocks (peridotite-biotite-pyroxenite series) represented only by ejected blocks and inclusions. New rock names katungite, murambite, and ugandite. Twenty-six detailed chemical yses are given. The rocks are compared in detail with those from r areas; their origin is discussed and evidence is adduced of trans- n [Min. Mag. 24-367, 408].

L. J. S.

MES (Arthur). *The petrology of the katungite.* Geol. Mag. London, 1937, vol. 74, pp. 200-219, 1 pl.

The lava of the extinct volcano of Katunga in SW. Uganda is named ngite. It is a potash-rich (K_2O 2·65-4·09%) olivine-melilitite, with ocrysts of olivine (α 1·660, $2V$ 90°) and melilite (ω 1·660, ϵ 1·655), ther with perovskite, magnetite, and rarely leucite, in a cryptocrystall- or glassy groundmass. Vesicles contain phillipsite, ashcroftite? and β about 1·535, γ 1·544), and natrolite. Detailed analyses of the s show SiO_2 35·37-37·93, Al_2O_3 6·50-6·83, Fe_2O_3 6·81-9·68, FeO

2·70–5·00, MgO 11·67–14·54, CaO 15·23–16·79, Na₂O 0·88–1·56 %, &c.
Remarks are added on rock names and on petrogenesis. L. J. S.

[NIKOLAEV (V. A.)] Николаев (В. А.). Щелочные породы р. Каинды в Таласском Алатау.—NIKOLAEV (V. A.). *The alkali rocks of the Kaindy river in the Thalass Alatau range.* Труд. Центр. Научно-Иссл. Геол.-Разв. Инст. (Trans. Central Geol. Prospl. Inst. U.S.S.R.) 1935, no. 11, 112 pp., 1 map, 8 pl., 4 text-figs. (Russian with English summary.)

A description is given of an igneous stock 3·5 sq. km. intruded into the upper Devonian lower Carboniferous limestones 160 km. NE. of Tashkent. The plutonic complex, of a fluidal-banded structure, is composed of biotite-pyroxenite (anal. I), shonkinite (anal. II), and monzonite-gabbro (anal. III). The associated dikes include biotite-pyroxenite, shonkinite, fergusite (anal. IV), nepheline-syenite (anal. V), and quartz-syenite (anal. VI). Pseudoleucite (orthoclase-nepheline) is present in certain varieties of biotite-pyroxenite and is very abundant in fergusite (over 50 %), which also contains dactylitic (myrmekitic) intergrowths of orthoclase and nepheline. Fergusite is compared to borolanite from the type locality. The exocontact rocks are marble and calc-silicate hornfels. The mechanism of emplacement, assimilation, and differentiation are discussed, and altogether 9 chemical and 13 modal analyses are given.

	I.	II.	III.	IV.	V.	VI.
SiO ₂	40·83	46·40	50·48	48·33	54·87	62·0
TiO ₂	1·35	1·15	0·88	1·10	0·59	0·4
Al ₂ O ₃	5·53	11·51	17·72	17·46	20·72	17·6
Fe ₂ O ₃	10·39	6·15	5·26	4·81	3·10	2·7
FeO	6·67	5·60	3·73	3·47	1·44	1·4
MnO	0·24	0·34	0·25	0·12	0·22	0·2
MgO	13·28	7·65	5·30	3·10	0·45	1·2
CaO	17·22	12·25	10·02	6·67	3·28	4·6
BaO	0·34	0·58	0·23	0·71	0·34	0·3
Na ₂ O	0·54	1·89	2·62	2·60	6·02	4·0
K ₂ O	1·50	4·60	2·72	9·45	8·52	5·0
P ₂ O ₅	1·18	1·18	0·61	0·72	0·11	0·2
CO ₂	—	—	—	0·69	0·07	0·0
Cl	—	—	0·03	n.d.	0·02	—
H ₂ O+	0·60	0·64	0·49	0·77	0·59	0·3
H ₂ O—	0·08	0·14	0·15	0·07	0·14	0·2
	99·75	100·08	100·49	100·07	100·48	100·5
Sp. gr.	3·28	3·06	2·95	2·83	2·68	2·7

S. I. T.

KVIN (A. V.)] Москвин (А. В.). Щелочные породы верховьев реки Исфайрам. *Alkaline rocks from the upper part of Isfairyam river.* Труд. Петр. Инст. Акад. Наук СССР (Trav. Inst. Pétrogr. Acad. Sci. U.R.S.S.), 1934, no. 6, pp. 251–263, 2 figs. (Russian with English summary.)

This is a preliminary account of an alkali-granite and alkali-syenite intrusion into Palaeozoic limestone and schist in the central part of the range, Turkestan. Five analyses are given. S. I. T.

KRITZKY (A. N.)] Заварицкий (А. Н.). Щелочные горные породы Ишима.—ZAVARITSKY (A. N.). *The alkaline rocks of the Ishim river (west Siberia).* Труд. Петр. Инст. Акад. Наук СССР (Trav. Inst. Pétrogr. Acad. Sci. U.R.S.S.), 1936, no. 7–8, pp. 47–102, 1 pl., 22 text-figs. (Russian with English summary.)

The following alkali rocks are described: cancrinite-melanite-syenite with primary calcite, scapolite-hornblende-syenite, porphyritic corundum-syenite-microsyenite, fluorite-melanite-nepheline-syenite, aegirine-augite-nepheline-syenite, pyroxene-syenite or monzonite, essexite, clinopyroxene, epilcucite-porphyry [M.A. 6–418]. Certain metamorphic rocks are also described. The field relations of the intrusive rocks are not clear, but apparently they are associated with the acid intrusive rocks of the region. S. I. T.

ELLIE (D.). *Leucite-basanite from East Lothian.* Geol. Mag. London, 1936, vol. 73, pp. 16–19.

A rock of a volcanic plug and dike on the coast near Tantallon, previously referred to as 'limburgite', is a leucite-basanite with pyritic crystals of augite, olivine, and hornblende in a groundmass of plagioclase, augite, magnetite, analcime, leucite, and glass. The leucite crystals (0·1 mm.) are largely altered to analcime. L. J. S.

RUM (J. A.). *Analcite rock from Cloudy Bay, Marlborough, New Zealand.* New Zealand Journ. Sci. Techn., 1936, vol. 18, pp. 120–123, 2 figs.

Samples of an analcime-tinguaite, resembling huronite, contain abundant phenocrysts of analcime pseudomorphous after nepheline, and of aegirine and aegirine-augite, in a groundmass consisting of felted needles of aegirine-augite enclosed in analcime. L. J. S.

COULSON (A. L.). *A soda-granite suite in the North-West Frontier Province.* Proc. Nat. Inst. Sci. India, 1936, vol. 2, pp. 103–111.

Chemical analyses are given of biotite-aegirine-arfvedsonite-gneiss (soda-granite) from Khyber and of soda-porphry containing aegirine and arfvedsonite from Peshawar. Alkali rocks have not been previously recorded from the Himalayan region.

L. J. S.

[KUZNETSOVA (E. V.)] Кузнецова (Е. В.). Схема классификации докембрийских гранитных пегматитов горной Балкарии.—

KUZNETSOVA (El. V.). *Classification (scheme) of the pre-Cambrian pegmatites of the High Balkaria.* Зап. Всеросс. Мин. Общ. (Mémoires Soc. Russe Min.), 1936, ser. 2, vol. 65, pp. 385–417, 20 figs. (Russian with English summary.)

Pegmatite veins are found along the margin and in the roof of a granodiorite intrusion of High Balkaria (central Caucasus). In order of abundance the minerals of the pegmatites are: quartz, microcline, plagioclase, muscovite, biotite, tourmaline, garnet (almandine, andradite), hornblende, magnetite, apatite, beryl, columbite, tantalite, chlorite sulphides (pyrite, pyrrhotine, arsenopyrite). The pegmatites are classified according to the scheme of A. E. Fersman into: (1) Pegmatites of pure descent; (2) Pegmatites of hybrid descent (contact pegmatites, migmatite pegmatites).

S. I. T.

[ALTHAUSEN (M. N.)] Альтгаузен (М. Н.). Месторождение Колотовки Мамско-Витимского слюдоносного района.—

ALTHAUSEN (M. N.). *The Kolotovka deposit of the Mama-Vitim mica-bearing region.* Труды Всесоюзн. Научно-Иссл. Инст. Мин. Сырья (Trans. All-Union Scientific Research Inst. Econ. Min.), 1935, no. 86, 68 pp., 1 pl., 37 text-figs. (Russian with English summary.)

Numerous pegmatite veins intrude into the metamorphic rocks of the region. The latter include kyanite-gneiss, quartzite, and crystalline limestone. The following types of pegmatite are distinguished: (1) medium-grained graphic microcline-pegmatite; (2) medium-grained mica-plagioclase-pegmatite; (3) gigantic pegmatite in which large crystals of muscovite, often intergrown with biotite, are found. The last type usually occurs along the contact with gneiss. Shot-like spheroids of graphite were found enclosed in quartz in one of the pegmatites. Several analyses of pegmatites and metamorphic rocks are given, and the paragenesis of the minerals discussed.

S. I. T.

LER (G. D.)] Аллер (Г. Д.). Адун-Чолонские граниты.—ALLER (H.). *The granites of Adun-Cholon range.* Зап. Всеросс. Мин. Общ. (Mém. Soc. Russe Min.), 1936, ser. 2, vol. 65, pp. 289–314, 2 pls., 6 text-figs. (Russian with English summary.)

detailed petrological account is given of the granites and the associated pegmatite and aplite veins of this well-known mineral locality. A scheme of the order of crystallization is suggested, together with a genetic classification of pegmatites. Several analyses of granites and schists are given.

S. I. T.

KOVÁ (Božena). *O množství živce a křemene v pegmatitech.* [The quantity of felspar and quartz in pegmatites.] Příroda, Brno, 1936, vol. 29, pp. 233–236.

Microscopic measurements of felspar and quartz in pegmatites, from Raž near Písek, Bohemia, and from Dolní Bory, Cyrilov, and Rožná Moravia, confirm that there is no quantity relation corresponding to eutectic. In twelve samples, quartz ranged from 21·6 to 34·4 % weight.

F. S.

PAR (Jan Václav). *Stručný nástin mineralogie a geochemie řičanské žuly.* [Brief note on the mineralogy and geochemistry of the Ríčany granite.] Věda Přírodní, 1936, vol. 17, pp. 168–171, 1 map. Pegmatite veins intruded into contact-metamorphosed schists show symmetrical arrangement—felspars, tourmaline, quartz—with a tendency to monomineralic composition. The paragenesis of chemical elements is characterized by the accumulation of potassium towards end of the magmatic processes. Cassiterite and mispickel are rarely crystallized constituents of the pegmatites.

F. S.

GA-PONDAL (I.) & FRAGA-PADIN (M^a. de la E.). *Químismo de la pegmatita litinífera de Goyás (Lalin).* Anal. Soc. Española Fís. Quím., 1934, vol. 32, pp. 1011–1020, 1 fig.

Analyses of the pegmatite (I), spodumene (II), and muscovite (III, MgO 0·06) show that in the minerals lithium is present only in II. Spodumene forms crystals up to 10 cm. long and is in part altered.

SiO ₂ .	Al ₂ O ₃ .	Fe ₂ O ₃ .	FeO.	MnO.	CaO.	Li ₂ O.	Na ₂ O.	K ₂ O.	H ₂ O.
72·16	17·68	0·30	0·21	0·04	0·09	1·20	6·56	0·92	0·86
62·65	28·20	0·20	0·14	0·11	nil	5·55	1·22	1·21	0·75
45·13	37·50	1·25	0·10	0·02	nil	nil	1·21	10·54	4·36

L. J. S.

CAMPBELL (Ian). *Types of pegmatites in the Archaean at Grand Canyon, Arizona.* Amer. Min., 1937, vol. 22, pp. 436-445, 2 figs.

Three types of granite pegmatites are recognized. I, Intrusive pegmatites, consisting almost entirely of quartz and microcline. II, Replacement pegmatites, formed by 'pegmatization' (analogous to granitization), with albite as the chief felspar. III, Felspathized quartz veins, showing replacement of quartz by microcline. These different types are not easily distinguished, unless seen in an early stage of their formation.

L. J. S.

BJØRLYKKE (Harald). *Mineral paragenesis of some granite pegmatites near Kragerø, southern Norway.* Norsk Geol. Tidsskr., 1937, vol. 17, pp. 1-16, 6 figs.

The minerals found in the felspar quarries are noted. The pegmatites of this district differ from those of other parts of Norway in being rich in calcium, with the minerals calcite, hellandite, sphene, betafite, and apatite, and also in an abundance of tourmaline and scarcity of beryl [M.A. 6-214, 231].

L. J. S.

BJØRLYKKE (Harald). *The granite pegmatites of southern Norway.* Amer. Min., 1937, vol. 22, pp. 241-255, 5 figs.

Summary of an earlier paper [M.A. 6-214].

L. J. S.

HORWOOD (H. C.). *Granitization in the Cross Lake Region, Manitoba.* Trans. Roy. Soc. Canada, 1936, ser. 3, vol. 30, sect. 4, pp. 99-117, 7 figs.

At contact with intruded granite, andesite shows a gradual transition into granodiorite-gneiss and arkose into granite-gneiss. In the andesite there is a gain of quartz, potash-felspar, and biotite, and loss of hornblende; in the arkose potash-felspar replaces quartz. The chemical composition of the rocks (calculated from micrometric analysis of thin sections) shows in the andesite gain of SiO_2 and K_2O , and loss of Fe_2O_3 , FeO , MgO ; and in the arkose gain of Al_2O_3 and K_2O with loss of SiO_2 . These changes were effected by dilute alkaline solutions containing SiO_2 , Al_2O_3 , and K_2O emanating from the granite at about 600° C.

L. J. S.

BARTH (Tom. F. W.). *Structural and petrologic studies in Dutchess County, New York.* Bull. Geol. Soc. Amer., 1936, vol. 47, pp. 775-850, 4 pls., 8 text-figs.

Palaeozoic argillaceous sediments show progressive metamorphism in which muscovite-slate, kyanite-schist, and fibrolite-gneiss facies are

guished. Several chemical analyses are given of the rocks and of minerals isolated from them. A four-fold genetic classification of includes igneous, syntectic, metamorphic, and sedimentary. Syntectic rocks are those formed by stewing of previously solidified rocks in s of magmatic and/or palingenic origin; e.g. felspathized schists and pelite.

L. J. S.

U (Kazuo) [1899-1937]. *A remarkable association of several thermal and stress minerals in the Yu hsi kou iron-bearing district.* Mem. Ryojun Coll. Engineering, 1935, vol. 8, no. 8, pp. 167-186, 2 pls., 13 text-figs.

The intrusion of tourmaline-granite into gneisses and crystalline limestone has produced hornfelses of various types with garnet, hypersthene, phyllite, grunerite, xanthophyllite, olivine, quartz, and epidote. Hypersthene, as crystals up to 20 cm. long, has sp. gr. 3.81, α 1.7549, β 1.7727, γ 1.7727, 2V 63-76°, negative [cf. Min. Mag. 24-527]. The iron consists of magnetite disseminated in crystalline limestone and in hornfelses. The different mineral associations are considered.

L. J. S.

SPEED (G. E.) & COOMBS (Howard A.). *Replacement breccias of the Lower Keechelus.* Amer. Journ. Sci., 1937, ser. 5, vol. 34, pp. 12-23, 3 figs.

The rocks of this formation of Tertiary age in the Cascade Ranges of Washington have been thought to be of pyroclastic origin. They show brown fragments of sedimentary rock in a greyish-green matrix. It also contains remnants of sedimentary material. Analyses of two portions show an increase of SiO₂ and K₂O and a decrease of FeO, MgO, and CaO in the matrix, the composition of which is similar to that of a dacite. It is suggested that sandy shales invaded granodiorite and intensely folded underwent recrystallization by a process of additive hydrothermal metamorphism.

L. J. S.

LIPS (A. H.) & HESS (H. H.). *Metamorphic differentiation at contacts between serpentinite and siliceous rocks.* Amer. Min., 1936, vol. 21, pp. 333-362, 10 figs.

In the serpentine belt extending from Vermont into southern Quebec the country-rock is usually quartz-mica-schist or phyllite. At the contact in the lower temperature type of metamorphic differentiation, there is a zone of talc on the serpentine side with magnesite veins in the serpentinite, and a zone of chlorite on the country-rock side. In a higher

temperature type these zones consist respectively of actinolite and biotite; but this may be complicated by lower temperature type being superposed. Several rock analyses are given, and the transference of material from one rock to another as well as the addition of material by hydrothermal solutions are discussed. Similar changes in other areas e.g. Shetland [Min. Mag. 23-519], are compared.

L. J. S.

MOUNTAIN (Edgar D.). *Syntectic phenomena in Karroo dolerite Coedmore quarries, Durban.* Trans. Geol. Soc. South Africa, 1933, vol. 38 (for 1935), pp. 93-112, 2 pls., 1 text-fig.

A sheet of dolerite intrusive in Table Mountain Sandstone has enclosed masses of the latter and has taken a certain amount of silica into solution. The incorporation of sediment has influenced the plagioclase of the dolerite in several ways. When silicification took place before the crystallization of labradorite was finished, the latter was replaced by a more acid plagioclase; but when the labradorite had completed its crystallization it was replaced by micropegmatite or even by chlorite. The modified dolerite ranges in composition from quartz-dolerite to augite granophyre.

S. J. S.

[ALESHKOV (A. N.)] Алешков (А. Н.). Геологический очерк района горы Неройки.—ALESHKOV (A. N.). *Short geological description of the Neroika Mnt region* [sic]. Приполлярный Урал [Subpolar Ural Sov. Изуч. Произв. Сил СССР и Петрогр. Инст. им. Ф. К. Левинсон-Лессинга [Council Research Econ. Res. U.S.S.R. Petrogr. Inst. Levinson-Lessing], 1937, Ural Ser., no. 6, pp. 3-52, 22 figs. (Russian with English summary.)

[ALESHKOV (A. N.)] Алешков (А. Н.). Геологический очерк района Сура-Из.—ALECHKOV [sic] (A. N.). *Short geological description of the Sura-Iz Mnt region.* Ibid., pp. 57-86, 19 figs. (Russian with English summary.)

[LAEMMLEIN (G. G.)] Леммлейн (Г. Г.). Кристаллографическое исследование кварца с горы Сура-Из.—LAEMMLEIN (G.). *Crystallographic investigation of quartz from Sura-Iz Mnt.* Ibid., pp. 87-94, 1 fig. (Russian with English summary.)

The geological structure of these two adjoining districts is similar. The metamorphic rocks (phyllite, micaite, listwanite, marble, and quartzite) were injected by spilitic magma mainly represented by green schists. This was followed by the injection of granitic magma, which according to the author, was produced by the anatexis of para-schists.

itic magma interacting with basic rocks gave rise to a series of rocks including gabbro-diorite, diorite, and syenite. Among the rock types are quartzolite, a palingenic quartzite and calcite, as an intrusive rock cutting gabbro. Rock-crystal is found in scree and it is probably associated with the injection of quartz-silicate into the green schists. One of the quartz crystals from the Šika region is 120×75 cm. and weighs 1700 kg. Quartz crystals are encrusted with chlorite and contain inclusions of green schist, ilite, brookite, magnetite, rutile, sphene, and anatase. Some show hints of tabular calcite crystals, and this together with a singular orientation of quartz and calcite suggests their paragenesis from late magmatic solutions.

S. I. T.

KSDALE (Julian D.). *The Shonkin Sag laccolith*. Amer. Journ. Sci., 1937, ser. 5, vol. 33, pp. 321–359, 1 pl., 10 text-figs.

This layered laccolith in the Highwood Mts. of Montana, which has been regarded as a classic example of differentiation in place from a single intrusion, is now described as a compound laccolith formed by successive intrusions of shonkinite, syenite, and pegmatite. L. J. S.

SHALL (P.). *Acid rocks of the Taupo-Rotorua volcanic district*. Trans. Roy. Soc., New Zealand, 1935, vol. 64, pp. 323–365, 9 pls., 1 text-fig. (sketch-map).

A further account of the ignimbrites [M.A. 5–294], with descriptions of three main types, one of which has long been known under the name 'Wilsonite'. Published analyses are tabulated with one new analysis of 'tridymite-rhyolite' from Waihi. The analysed rocks have the composition of rhyolites with rather high lime. Ignimbrites are defined as vitreous rocks of acid or perhaps intermediate composition which have been formed from material that has been ejected from orifices in the form of a multitude of highly incandescent particles which were mainly of minute size'. The temperature of the material when deposited seems to have been at least 1200° C., but below 1470° C. Some heating tests on various obsidians are described. A classification into 'pulverulites, sculites, and lapidites' is based on structure, and in each group there may be vitreous, radial, plumose, or pectinate phases. W. C. S.

SHALL (P.). *Geology of Mayor Island*. Trans. Roy. Soc. New Zealand, 1936, vol. 66, pp. 337–345, 1 fig.

Activities in comendites contain small crystals of aegirine and riebeckite [M.A. 5–295]. Riebeckite as long blades up to 1 cm., with (010) narrow

and prisms relatively wide, has extinction $2\cdot5^\circ$, pleochroism α dark-blue, β pale-blue, γ brownish-yellow. Some cavities and joint-planes contain opal with a fine play of colour with scarlet fire. Compact comedendites at Opo bay contain also crystals of aenigmatite, and in the groundmass mossy growths of the variety cossyrite. Tuhalite [M.A. 6-536] occurs in situ in rocks in the crater wall. Some of the comedendites develop spherulitic and banded structures. Two new analyses of comedendite are given.

W. C. S.

SANDFORD (K. S.). *Geological observations on the north-west frontiers of the Anglo-Egyptian Sudan and the adjoining part of the southern Libyan Desert.* Quart. Journ. Geol. Soc. London, 1935, vol. 91, pp. 323-381, 7 pls. (map), 11 text-figs.

Two areas of lavas and a number of scattered outcrops of intrusive rocks, post-Nubian sandstone in age, are described. One isolated dike cutting Nubian sandstone, west of Laqiya ($19^\circ 53' N.$, $27^\circ 20' E.$) is described as a quartz-tuff. It consists of devitrified glass with jagged fragments of quartz. Sandstone-walls of craters in the volcanic field between 'Uweinat and the Gilf Kebir show a selvage consisting of isotropic black or brown glassy material cementing shattered quartz grains. Similar material glazes the walls of remarkable cavities which traverse the surface of the sandstone near the craters. The cavities are usually empty, but some contain fragments of lithomarge or of decomposed trachyte. The cavities are fissures up which molten rock may have ascended and subsided before cooling. The cavity walls are highly silicified sandstone and stand up 10, or sometimes as much as 30, feet above the surface. Photomicrographs are given of one of the specimens from the selvage on a crater wall, also of basanite, phonolite, trachyte, soda-trachyte, and trachytic phonolite.

W. C. S.

SANDFORD (K. S.). *Extinct volcanoes and associated intrusions in the Libyan Desert.* Trans. Roy. Geol. Soc. Cornwall, 1935, vol. 16, pp. 331-358, 4 pls., 2 text-figs.

Further descriptions of the volcanic fields described in the paper abstracted above. A photomicrograph is now given of the 'quartz-tuff' dike west of Laqiya, also of a monchiquite from a loose block at the foot of Sandara Hill some 100 km. SW. of 'Uweinat. This paper describes concentrations of heavy black material rich in manganese both as irregular lumps in the volcanic rocks near Sandara Hill, and also at the contact

volcanic rocks with the glazed sandstone walls described in the
ed abstract. W. C. S.

WER (H. I.). *Symplectite-bearing nodules in the Ardgour marble, Argyllshire*. Geol. Mag. London, 1936, vol. 73, pp. 448-468, 6 figs.
odules, probably representing a band of impure limestone in the
d marble, consist of idocrase, scapolite, garnet, augite, bytownite,
oclase, clinozoisite, calcite, &c. (24 minerals are noted), forming
olectic intergrowths and with indications of various mineral replace-
ts. Analyses by N. Sahlbom are given of: I, idocrase, also F 1·08 %,
tr. 3·42, ω 1·724, ϵ 1·715; II, pink pyroxene, sp. gr. 3·42, α 1·694,
 γ 1·699, γ 1·721, γ : c = 43 $\frac{1}{2}$ °, 2V 59°, positive.

SiO_2	TiO_2	Al_2O_3	Fe_2O_3	FeO	MnO	MgO	CaO	Na_2O	K_2O	H_2O
36·53	3·75	15·79	2·05	2·11	0·11	2·61	35·70	0·38	0·21	0·70
46·32	1·20	6·51	1·48	8·45	0·15	10·45	24·10	0·31	0·59	0·44

L. J. S.

KEY (C. E.). *Enderbite, a new member of the charnockite series*. Geol. Mag. London, 1936, vol. 73, pp. 312-316.

ocks from Enderby Land, Antarctica, include acid, basic, and ultra-
e members of the charnockite series, and also a related type, called
erbite, in which plagioclase (as antiperthite) is the predominant
ar, rather than microcline (as microperthite). This rock has a
um-grained granitic texture with faint foliation, and is composed
uartz 42·5, antiperthite 53 (rods of orthoclase 4·5, in plates of
esine 48·5), hypersthene 3·0, and magnetite 1·0 %. Analysis shows
K₂O and high CaO: SiO₂ 75·50, TiO₂ 0·08, Al₂O₃ 13·92, Fe₂O₃ 0·48,
1·17, MgO 0·52, CaO 3·54, Na₂O 3·64, K₂O 0·74, H₂O+ 0·15,
- 0·30 = 100·04, sp. gr. 2·67. Birkremite (of C. F. Kolderup, 1903)
a Norway is a charnockite.

L. J. S.

LEBEDEV (A. P.)] Лебедев (А. П.). Эпидотовый диорит из восточной Сибири.—LEBEDEV (A. P.). *Epidote-diorite from the east Siberia*. Труд. Петр. Инст. Акад. Наук СССР (Trav. Inst. Pétrogr. Acad. Sci. U.R.S.S.), 1936, no. 7-8, pp. 105-112, 2 figs. (Russian with English summary.)

orite with primary epidote is found in the basin of the river Niukzha, tributary of the Olekma. The analysis of the diorite by M. Ermolaeva : SiO₂ 59·86, TiO₂ 0·76, Al₂O₃ 19·73, Fe₂O₃ 1·60, FeO 2·88, MnO MgO 1·61, CaO 6·46, Na₂O 5·02, K₂O 1·25, H₂O+ 0·79, H₂O- = '100·19'. Epidote occurs in the form of prismatic crystals, up to 5·2 mm. long, and usually enclosed in biotite; α 1·730, γ 1·769,

$\gamma - \alpha$ 0·043, 2V 75°, negative, strong dispersion. Orthite (pleochroic 2V 87°, negative) is sometimes found enclosed in epidote and optically continuous with it.

S. I. T.

[Rozhkov (B. N.) & Moore (G. G.)] Рожков (Б. Н.) и Мор (Г. Г.).

Граниты Анабарского докембрия и связанные с ними проявления металлоносности.—Rozhkov (B. N.) & Moore (G. G.). *Anabar pre-Cambrian granites and the phenomena of ore concentration connected therewith.* Изв. Акад. Наук СССР, Отд. Мат. Ест. Наук, Сер. Геол. (Bull. Acad. Sci. U.R.S.S., Cl. Sci. Math. Nat., Sér. Géol.), 1936, pp. 729–773, 1 fig. (Russian with English summary.)

Anabar (NE. Siberia) schists are injected by acid and basic magmas. The granites occur as stocks, sills, and migmatite injection complexes. The normal type of granite with microcline, near the contact with schists, especially in migmatite complexes, is transformed into plagioclase-, amphibole-, and hypersthene-granites. The exo-contact zone shows evidence of potash transfusion and an abundant development of potash-felspar and biotite. Pegmatite, aplite, and quartz veins are well represented, and some veins carry numerous ore-minerals, such as auriferous sulphides, monazite, cassiterite, &c. Four analyses of rock are given.

S. I. T.

[Ziv (E. F.) & Timofeev (V. D.)] Зив (Е. Ф.) и Тимофеев (В. Д.).

Шеелитоносность скарнов восточных отрогов Кузнецкого Алатау.—Ziv (E. F.) & Timofeev (V. D.). *The scheelite in the "scarns" of the eastern branches of the Kuznetsk Ala-Tau.* Изв. Акад. Наук СССР, Отд. Мат. Ест. Наук, Сер. Геол. (Bull. Acad. Sci. U.R.S.S., Cl. Sci. Math. Nat., Sér. Géol.), 1936, pp. 869–911, 10 figs. (Russian with English summary.)

The igneous rocks of this region in Siberia (54° N., 90° E.), in order of their consolidation, are gabbro, diorite, granodiorite, pegmatite, and aplite veins. Contact metamorphism of the adjoining sediments gave rise to marble and hornfels. In the penummatolytic and hydrothermal magmatic stages metamorphism was followed by extensive metasomatism which gave rise to skarn rock. According to the dominant mineral, the skarns are subdivided into pyroxene, garnet, magnetite, calcite, quartz, epidote, serpentine, diopside, scapolite, wollastonite, and amphibole skarns. Scheelite is usually associated with garnet and pyroxene-garnet skarn. Three analyses of scheelite, two of diopside, and several analyses of rocks are given.

S. I. T.

LEMS (H. W. V.). *Contribution to the petrology of the crystalline schists of western central Celebes (Netherlands East Indies)*. Proefschrift Univ. Amsterdam. Amsterdam (J. F. Duwaer), 1937, 147 pp. (English), 8 pp. (Dutch summary), 1 pl., 11 text-figs.

rocks collected SW. of Poso include crystalline limestones (some containing lawsonite), phyllites, quartzites, and various schists, including chlorite-quartz-mica-schist, glaucophane-lawsonite-quartz-chlorite-schist, act-epidote-glaucophane-schist, lawsonite-orthite-schist, crossite-quartz-rock, &c. Chemical analyses are given of calcareous phyllite and glaucophane-lawsonite-schist.

L. J. S.

OZEROV (K. N.) & BYKHOVER (N. A.)] Озеров (К. Н.) и Быховер (Н. А.). Месторождения корунда и кианита Верхне-Тимптонского района Якутской АССР.—OZEROV (K. N.) & BYKHOVER (N. A.). *Corundum and cyanite deposits of the Verkhne-Timpton district of the Yakutian Autonomous Soviet Socialist Republic*. Труд. Центр. Научно-Иссл. Геол.-Разв. Инст. (Trans. Centr. Geol. Prospr. Inst.), 1936, no. 82, 106 pp., 6 pl., 1 map. (Russian with English summary.)

Corundum-kyanite-rock occurs in the form of flat lenses among mica-like diaphthorite mylonitic epigneisses, named 'gneissoids' by Korzhinsky. Corundum, usually ruby-red and rarely blue in colour, occurs as tabular crystals. Kyanite is grey or green in colour. Green variety is called chrome-kyanite; its analysis by E. V. Vavitch gave SiO_2 33.76, TiO_2 1.10, Al_2O_3 62.46, Cr_2O_3 1.81, Fe_2O_3 0.29 = 99.76 with α 1.718, γ 1.734, $2V$ 82°, negative. Other minerals present in this rock are muscovite, chlorite, margarite, tourmaline, diasporite, &c. The rock probably represents a metamorphosed gneiss. Various schists, gneisses, and intrusive igneous rocks are also described and a number of analyses given.

S. I. T.

KUPLETSKY (B. M.)] Куплетский (Б. М.). Количественно-минералогический состав основных пород.—KUPLETSKY (B. M.). *Quantitative mineralogical composition of basic rock [sic]*. Труд. Петр. Инст. Акад. Наук СССР (Trav. Inst. Pétrogr. Acad. Sci. U.R.S.S.), 1936, no. 9, pp. 53–83, 13 figs. (Russian with English summary.) A statistical study of 690 modal analyses of basic and ultrabasic rocks provided the author with a rich material for a rational scheme of classification [cf. M.A. 6–313]. Variation curves for various components are constructed and discussed. Mesocratic gabbros, in their position, fall into the region of plagioclase-pyroxene cotectic, and

this apparently supports Vogt's idea that these rocks are anchi-eutectic. Two lines of differentiation of basic magma are suggested: (1) gabbro-diabasic, and (2) norito-anorthositic.

S. I. T.

[USTIEV (E. K.)] Устинев (Е. К.). Тридимитовый дацит с Кельском плато в Центральном Кавказе.—*On the tridymite-dacite from volcanic region of Keli in the middle Caucasus.* Труд. Петр. Инст. Акад. Наук СССР (Trav. Inst. Pétrogr. Acad. Sci. U.R.S.S.), 1935, no. 6, pp. 159–164, 2 figs. (Russian with English summary.)

Tridymite-dacite, containing on an average 5% of tridymite in vesicles, is found in the form of red taxitic bands in a grey dacite lava flow. Tridymite occurs as fairly large crystals with pearly lustre; $\gamma - \alpha$ 0·002, sign positive, elongation negative, optic axial angle small. A small amount of cristobalite with α 1·4824, γ 1·4876, is also present. Analysis by V. A. Moleva of tridymite-dacite gave SiO_2 66·30, TiO_2 0·66, Al_2O_3 17·32, Fe_2O_3 1·12, FeO 2·25, MgO 1·69, CaO 3·70, Na_2O 4·83, K_2O 1·77, ign. loss 0·25 = 99·82. A partial analysis of the grey dacite gave SiO_2 66·82. The formation of tridymite and the red colouring of the rock (due to the inclusion of minute flakes of haematite in the glass base) is attributed to the action of magmatic volatiles.

S. I. T.

[BELYANKIN (D. S.) & EREMEEV (V. P.)] Белянкин (Д. С.) и Еремеев (В. П.). Вулканические стекла Аджаристана.—*BELJANKIN (D. S.) and EREMEYEV (V. P.). Volcanic glasses of Adjariastan.* Труд. Петр. Инст. Акад. Наук СССР (Trav. Inst. Pétrogr. Acad. Sci. U.R.S.S.), 1935, no. 5, pp. 153–168, 2 pls., 2 text-figs. (Russian with English summary.)

Volcanic glasses are found associated with the Eocene porphyritic lavas of Adzharistan (the basin of the river Chorokh, near Batur Transcaucasia). They are almost black in colour with conchoidal fracture and have the appearance of obsidian. The texture is perlitic, fluidal, or ataxitic. A small amount of crystalline minerals (andesine, clinopyroxene, and magnetite) is present in almost all glasses. The analysis of a homogeneous glass from 3 km. west of Erge by V. A. Moleva gave SiO_2 57·20, TiO_2 0·78, Al_2O_3 15·61, Fe_2O_3 2·12, FeO 3·8, MnO 0·14, MgO 1·21, CaO 6·40, Na_2O 1·84, K_2O 1·14, $\text{H}_2\text{O} + 7\%$, $\text{H}_2\text{O} - 2\cdot50 = 99\cdot87$; sp. gr. 2·3648, n 1·533. Chemically this glass corresponds to andesito-dacite. The determination of water in other samples of glass shows a variation from 6 to 12% and of refractive indices from 1·558 to 1·508.

S. I. T.

YANKIN (D. S.) & IVANOVA (V. P.)] Белянкин (Д. С.) и Иванова И. П.). Термооптический анализ вулканического стекла.—
BELJANKIN (D. S.) & IVANOVA (V. P.). *Thermo-optical analysis of volcanic glass from Adzharistan.* Труд. Петр. Инст. Акад. Наук СССР (Trav. Inst. Pétrogr. Acad. Sci. U.R.S.S.), 1934, no. 6, pp. 381–392, 10 figs. (Russian with English summary.)

The volcanic glass from Erge (preceding abstract) was subjected to thermal study. The heating curve suggests that one part of water—'absorbent water'—is given out below 100° C., and is wholly readSORBED at exposure of the sample in air. The 'constitutional water' is given between 100° and 500° C. and the rehydration is only partial. The loss of water is accompanied by the solidification of glass, as judged by refractive index curve. Two exothermal breaks at 350° and 600° are attributed to the oxidation of ferrous iron in the glass. S. I. T.

OUSOVA (V. T.)] Белоусова (В. Т.). Трапы с реки Хатанги.—
BELOUSOV (T.). *The traps from r. Hatanga.* Труд. Петр. Инст. Акад. Наук СССР (Trav. Inst. Pétrogr. Acad. Sci. U.R.S.S.), 1936, no. 9, pp. 85–103. (Russian with English summary.)

The effusive and intrusive traps from the river Khatanga (Arctic Siberia) comprise the following types: olivine-dolerite with orthorhombic pyroxene, tholeiite, and basalt. Olivine, the crystallization of which preceded that of plagioclase, shows a variation of 2V 65–85°, negative, with marked zonal structure. Plagioclase ranges from labradorite to albite. Pyroxene is represented by enstatite, enstatite-augite and diopside-hedenbergite (2V 40–60°, negative, $\gamma:c = 30–46^\circ$, $\gamma - \alpha = 0.0249–0.0297$). The accessories are magnetite, biotite, chlorite, hornblende, apatite, talc, iddingsite. Four analyses of traps show their affinity to the Siberian traps and Arctic basalts and dolerites [M.A. 6–319].

S. I. T.

ZBERG (A. S.), TZVETKOV (A. I.), OSIPOV (M. V.), & RUDZIT (G. P.)] Гинзберг (А. С.), Цветков (А. И.), Осипов (М. В.) и Рудзит (Г. П.). Тулунский трапп как материал для каменного литья.—
GINSBERG (A. S.), ČVETCOV (A. J.), OSIPOV (M. W.), and RUDZIT (G. P.). *The Tulan traps as a material for stone smelting industry.* Труд. Петр. Инст. Акад. Наук СССР (Trav. Inst. Pétrogr. Acad. Sci. U.R.S.S.), 1936, no. 7–8, pp. 263–308, 20 figs. (Russian with English summary.)

The paper deals mainly with the petrurgy (science of cast rock material)

of Siberian traps and the properties of the cast ware. In the petrographical description of the raw material an account of Tulun dolerite is given [M.A. 6-317]. It consists of labradorite (An_{60}), pyroxene, iron ore, apatite, and olivine. Olivine is zonal with $2V\ 72^\circ$ (central) and $2V\ 62^\circ$ (outer), $\gamma - \alpha = 0.041$. The crystallization of olivine preceded and followed that of felspar, but olivine crystals enclosed in plagioclase have larger axial angles than those enclosing plagioclase laths. S. I. T.

[ZAVARITZKY (A. N.)] Заварицкий (А. Н.). Об атачите и некоторых новых данных по геологии месторождения г. Магнитной.

ZAWARICKIJ (ZAVARITSKY) (A.). *Sur l'atatchite et sur quelques nouvelles données concernant la géologie du gisement du mont Magnitnaya.* Изв. Акад. Наук СССР, Отд. Мат. Ест., Сер. Геол. (Bull. Acad. Sci. U.R.S.S., Math. Nat. Cl., Sér. Géol.), 1936, pp. 321-324 (Russian), pp. 324-327 (French).

Atatschite (from Atach ridge, Mt. Magnitnaya, Urals), previously described as an igneous rock (J. Morozewicz, 1901), is proved to be talc and conglomerate metasomatically altered under the influence of granitic magma. S. I. T.

Miscellaneous.

TERTSCH (H.). *Bemerkungen zur Frage der Verbreitung und zur Geometrie der Zwillingsbildungen.* Zeits. Krist., 1936, vol. 94, pp. 461-490, 2 figs.

Amongst 1400 mineral species twinning has been observed in 266 (19 %). The different types of twinning in each crystal system are enumerated. L. J. S.

TERTSCH (H.). *Beobachtungen an Orthoklas-Zwillingen nach dem Karlsbader Gesetz.* Zentr. Min., Abt. A, 1936, pp. 198-207, 4 figs.

Carlsbad twins of orthoclase of the habit with predominating (001) and ($\bar{1}01$) show a surface of combination in one plane parallel to (010)—pegmatite type. Those with predominating (001) and ($\bar{2}01$) show a stepping in the combination surface—rock type. L. J. S.

[BUSHINSKY (G. I.)] Бушинский (Г. И.). К вопросу о генезисе флюорита в осадочных породах.—BUŠINSKY (G. I.). *On the genesis of fluorite in sedimentary rocks.* Изв. Акад. Наук СССР, Отд. Мат. Ест. Наук, Сер. Геол. (Bull. Acad. Sci. U.R.S.S., Cl. Sci. Math. Nat., Sér. Géol.), 1936, pp. 775-793, 3 figs. (Russian with English summary.)

From his study of the fluorite and ratovkite deposits in the Moscow

ict and a comparison between them and similar deposits elsewhere
uthor premises that, in its initial stage of precipitation from sea-
r, the bulk of fluorine was concentrated in phosphorite. He ques-
the alleged connexion of sedimentary fluorite with magmatic
ity, the frequent association of fluorite with dolomitic or red-
red sediments suggesting that it has been formed in relict seas
. 2-124, 179].

S. I. T.

AVDIN (V. F.)] Алявдин (В. Ф.). Исследование кристаллов
оловянного камня из Сохондинского оловянного месторождения
(Южное Забайкалье).—ALJAWDIN (W.). *Die Untersuchung der
Zinnsteinkristalle von der Sochondolagerstätte (Sud-Transbaikalien).*
Зап. Всеросс. Мин. Общ. (Mém. Soc. Russe Min.), 1936, ser. 2,
vol. 65, pp. 444-453, 1 fig. (Russian with German summary.)

this region cassiterite is found in quartz veins together with
lite, arsenopyrite, and molybdenite. On the cassiterite crystals
amidal faces are well developed, while diprismatic faces are cor-
l. The dominant faces are (231) and (230). Faces (13.17.6),
(2.11), and (10.52.15) were also observed.

S. I. T.

EVINSKY (P. N.)] Чирвинский (П. Н.). Ортит и его парагенезис
в кристаллических породах Кольского полуострова.—TSCHIR-
EVINSKY (P.). *Orthit und dessen Paragenesis in den kristallinen
Gesteinen des Halbinsel Kola.* Зап. Всеросс. Мин. Общ. (Mém.
Soc. Russe Min.), 1936, ser. 2, vol. 65, pp. 163-177. (Russian with
German summary.)

all quantities of orthite (average 0.39 wt. %) are found in the pre-
cambrian gneisses, granite, and pegmatite. It is apparently rich in
earths, and some orthite grains enclosed in biotite and hornblende
surrounded by pleochroic haloes. Genetically it is connected with
granite magma.

S. I. T.

KUPLETSKY (B. M.)] Куплетский (Б. М.). Кнопит в породах основной
магмы.—KUPLETSKIJ (B. M.). *Knopite in basic magma rocks.* Изв.
Акад. Наук СССР, Отд. Мат. Ест., Сер. Геол. (Bull. Acad. Sci.
U.R.S.S., Cl. Sci. Math. Nat., Sér. Géol.), 1936, p. 109 (Russian),
pp. 110-111 (English).

Knopite is a knopite-rich rock occurring as veins in a pre-Cambrian
gneiss near Afrikanda station on the Murmansk railway. Knopite
is in this rock as fine aggregates of steel-grey isotropic grains,

n 2·37; analysis by V. A. Egorov gave SiO_2 0·96, TiO_2 56·35, rare earths 2·23, Al_2O_3 0·24, Fe_2O_3 0·78, FeO 0·70, MnO 0·02, MgO 0·1, CaO 37·52, $\text{H}_2\text{O} + 0\cdot57$, $\text{H}_2\text{O} - 0\cdot16 = 99\cdot67$. Knopitite veins are associated with the veins and segregations of titanomagnetite and probably represent a late differentiate of the pyroxenite magma. Nepheline-bearing segregations are also present.

S. I. T.

SZENTPÉTERY (S. v.). *Titanomagnetithältige Gesteine der Vaskapugege vom Bükkgebirge in Ungarn*. Acta Chem. Min. Physica, Szeged, 1937, vol. 6, pp. 55–100, 7 pls.

Analyses of titanomagnetite from anorthosite-gabbro (I) and peridotite (II) correspond with $2\text{Fe}_2\text{O}_3 \cdot 7\text{FeO} \cdot 5\text{TiO}_2$ and $\text{Fe}_2\text{O}_3 \cdot 7\text{FeO} \cdot 6\text{TiO}_2$ respectively, after deducting MgAl_2O_4 and MgTiO_3 .

	SiO_2 .	TiO_2 .	Al_2O_3 .	Fe_2O_3 .	FeO .	MnO .	MgO .	CaO .	Total.	Sp.
I.	0·10	32·20	1·74	23·98	40·10	0·42	0·70	—	99·24	4·7
II.	2·54	41·30	--	11·49	38·60	1·01	3·18	1·53	99·65	4·6

V. Z.

PALACHE (Charles). *Chalcomenite from Bolivia*. Amer. Min., 1933, vol. 22, pp. 790–795, 2 figs.

— *Supplementary note on chalcomenite*. Ibid., p. 1123.

Small, transparent, blue crystals in cavities in blockite = penroseite [M.A. 6–490] from Hiaco mine, Colquechaca, are orthorhombic-longsphenoidal (not monoclinic) with $a:b:c = 0\cdot7325:1:0\cdot8077$. The unit cell, a 6·56, b 9·10, c 7·36 Å., contains $4(\text{CuSeO}_3 \cdot 2\text{H}_2\text{O})$; space-group $P2_12_12_1$. Sp. gr. 3·35, H. > 2. $\alpha = a$, $\gamma = b$, α 1·712, β 1·732, γ 1·722. $2V$ nearly zero, negative.

Measurements by the late H. Ungemach of chalcomenite from Sier de Umango, Argentina, gave $a:b:c = 0\cdot7274:1:0\cdot7998$, and confirm the sphenoidal symmetry; new crystal-forms are (112) and (121).

L. J. S.

MINGUZZI (Carlo). *Sulla presenza della portlandite fra i prodotti vesuviani*. Periodico Min. Roma, 1937, vol. 8, pp. 5–13.

Analysis of a yellow powder on lava from the crater of Vesuvius showed Al, K, Na sulphates with a surplus of CaO (10·44 %), Al, H₂O. Under the microscope, minute (0·005–0·03 mm.) hexagonal prisms = portlandite $\text{Ca}(\text{OH})_2$ [Min. Mag. 23–419] with ω 1·57, ϵ 1·54 were identified; together with aphthitalite, aluminite, and amorphous aluminium hydroxide.

L. J. S.